

CIVIL SERVICES EXAMINATION (MAINS) 2020

PHYSICS PAPER - II: QUANTUM MECHANICS

TUTORIAL SHEET: 1

Foundations of Quantum Mechanics

1. How Davisson - Germer electron diffraction experiment confirms the de - Broglie hypothesis of matter wave? *gun (e⁻) p = 50°, θ = 65° λ = 1.65 Å* (1995)

2. Using Heisenberg uncertainty principle, find the ground state energy and Bohr radius of hydrogen atom. *defn it → write expression → proof → make diagram of v₁₅ m* (1997)

3. Using the uncertainty principle $\Delta x \Delta p \sim \frac{\hbar}{2}$, estimate the minimum energy of a particle in a simple

harmonic potential $U = \frac{1}{2} kx^2$ *defn → particle as matter wave group* (1999)

4. Find the de - Broglie wavelength associated with an electron of energy (i) 10 eV and (ii) 10 MeV *λ = 1.22 nm (10 eV) & davisson germer experiment m₀c² = 0.51 MeV* (1999)

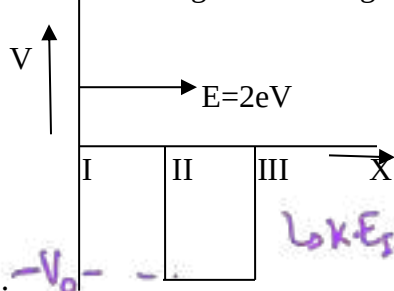
5. The momentum of an electron is 600 KeV/c. Determine its de Broglie wavelength and the phase and group velocities of its de Broglie waves. *λ = hc/p = hc/600 KeV = 2.06 × 10⁻¹² m* (2000)

6. A 200 eV increase in the energy of an electron changes its De Broglie wave length by a factor of two. Calculate the initial De Broglie wave length of the electron. *defn* (2002)

7. (i) Explain what do you understand by Heisenberg's uncertainty principle. Using this Principle, determine the energy of the ground state of a one dimensional simple harmonic Oscillator. *applicatⁿ → e⁻ waves + davisson expt. → Fraunhofer diffraction & specular reflection of neutral atoms if existence of atomic waves*

- (ii) An electron having an energy 2 eV is traveling in the region where V(x) varies as shown below:

Calculate the De Broglie wave length of the electron in region I, II and III. (2003)



8. (i) If x and P_x are two conjugate variables, show that

$$(\Delta x) (\Delta P_x) \geq |\langle [x, P_x] \rangle| / 2i$$

- (ii) An electron has a speed of 4×10^5 m/s, accurate to 0.01%. With what fundamental Accuracy can we locate the position of electron? *K.E._{II} = (2 + V₀) eV for non-relativistic case, λ = 1.22 nm / √(K.E. in eV)* (2004)

$$v \ll c \Rightarrow \text{non-relativistic case}$$

$$\Delta p \sim m_e \Delta v \quad |\Delta x| \sim \frac{\hbar}{2 m_e \Delta v}$$

9. Calculate the wave length of neutrons at a temperature of 20° C. (2004)

10. Calculate the de Broglie wavelength of an electron moving with a kinetic energy of 1 MeV. (2005)

11. Calculate the de Broglie wavelength of a thermal neutron at 27° C temperature. (2006)

12. (i) State and explain Heisenberg's uncertainty principle. Experimental data reveal that no electron in an atom has energy greater than 4 MeV. Assuming that the radius of a nucleus is 10^{-14} m, show using Heisenberg's uncertainty principle that an electron cannot exist inside the nucleus.

(ii) Calculate the de Broglie wavelength of thermal neutrons at 300 K. (2007)

13. Derive Bohr's angular momentum quantization condition in Bohr's atomic model from the concept of de Broglie waves. (2010)

14. Calculate the wavelength of De Broglie waves associated with electrons accelerated through a P.D. of 200 Volts. (2011)

15. Estimate the size of the hydrogen atom and the ground state energy from the uncertainty principle. (2011)

16. Use the uncertainty principle to estimate the ground state energy of a linear harmonic oscillator (2012)

17. In a series of experiments on the determination of the mass of a certain elementary particle, the results showed a variation of $\pm$ (plus-minus) 20 m(e), where m(e) is the electron mass. Estimate the lifetime of the particle. (2013)

18. Find the deBroglie wave length of (i) a neutron (ii) and electron moving with kinetic energy of 500 eV. (1 eV = 1.602×10^{-19} J) (2014)

19. The mean life of Lambda particle is 2.6×10^{-10} s. What will be the uncertainty in the Determination of its mass in eV? (2014)

20. Find the energy, momentum and wavelength of photon emitted by a hydrogen atom making a direct transition from an excited state with $n=10$ to the ground state. Also find the recoil speed of the hydrogen atom in this process. (2016)

21. An electron is confined to move between two rigid walls separated by 10^{-9} m. Compute the de Broglie wavelengths representing the first three allowed energy states of the electron and the corresponding energies. (2016)

22. A typical atomic radius is about 5×10^{-15} m and the energy of β -particle emitted from a nucleus is at most of the order of 1 MeV. Prove on the basis of uncertainty principle that the electrons are not present in nuclei. (2016)

→ If electrons were to be in nuclei, it would have been difficult to use photoelectric effect

which is basis of modern smoke detectors

23. Using uncertainty principle, calculate the size and energy of the ground state hydrogen atom. (2016)

define
 $a_0 = \frac{4\pi\epsilon_0\hbar^2}{m_e e^2} = 0.53 \text{ \AA}$

24. A beam 4.0 keV electrons from a source is incident on a target 50.0 cm away. Find the radius of the electron beam spot due to Heisenberg's uncertainty principle. (2017)

25. Estimate the de Broglie wavelength of the electron orbiting in the first excited state of the hydrogen atom. (2017)

26. Show that the mass and linear momentum of a quantum mechanical particle can be given by $m = h/(\lambda v)$ and $p = h/\lambda$, respectively, where h , λ and v are Planck's constant, wavelength and velocity of the particle, respectively. Comment on the wave-particle duality from these relations. (2019)

27. State and express mathematically the three uncertainty principles of Heisenberg. Highlight the physical significance of these principles in the development of Quantum Mechanics. (2019)

28. For a free quantum mechanical particle under the influence of a one-dimensional potential, show that the energy is quantized in discrete fashion. How do these energy values differ from those of a linear harmonic oscillator? (2019)

29. Using the uncertainty principle $\Delta x \Delta p \geq \hbar/2$, estimate the ground state energy of a harmonic oscillator. (2020)

Harmonic oscillator is ideal conception & simplest oscillator.

for 1-D, $V(x) = \frac{1}{2} m \omega^2 x^2$

Assuming a conservative system, $\frac{\partial E}{\partial x} \Big|_{x_0} = 0 \Rightarrow x_0 = \frac{\hbar}{\sqrt{2} m \omega}$

$E_0 = \frac{\hbar \omega}{2}$ is quantum mechanical result.

$V(x) = \frac{1}{2} m \omega^2 x^2$

TUTORIAL SHEET: 2

Schrodinger's Wave equation and applications

1. Write down Schrodinger's equation in one dimension and explain the significance of the eigen values and eigen functions of this equation.

Solve the Schrodinger's equation if the potential function V is given as

$$V(x) = V_0 \text{ (a constant) for } |x| < \frac{a}{2}.$$

$$a. = 0 \text{ for } |x| > \frac{a}{2}$$

State the boundary conditions you use along with their justification. (In the solution consider particles incident from one side only). If the total energy $E < V_0$, show that there is an important difference with classical mechanics. Explain, with the help of the uncertainty principle that $E < V_0$ does not mean that kinetic energy is negative.

2. Set up Schrodinger's equation for a one-dimensional harmonic Oscillator. Assuming that the solution is of the form $\psi = Ae^{-\alpha x^2} f(x)$, where $f(x)$ is a polynomial in x , find the energy Eigen values and the ground state Eigen function. Comment on the relation between the ground state energy and Heisenberg's uncertainty principle.

3. Set up Schrodinger's equation for a free particle confined in a cubical box and find the energy eigen values. What form will Pauli principle take for elections in such a box? Deduce the zero point energy if the length of the box be 10^{-10} metre and there be 10 electrons in it. (The coulomb interaction may be disregarded).

4. Using Schrodinger's equation show that a free particle in a box can have only discrete energy values.

5. Find the energy Eigen values of a one-dimensional harmonic oscillator whose Hamiltonian is given by
- $$H = \frac{p^2}{2m} + \frac{1}{2} kx^2$$

6. An election is confined in a one - dimensional box of $1 \text{ \AA}$ width. Draw the energy level diagram up to three energy states and also draw the corresponding normalized Eigen functions. Derive the expressions for Eigen functions and Eigen values used in the calculations. Show that Eigen functions are orthogonal.

7. An electron moving with energy E encounters one dimensional potential step as given below:

$$V(x) = 0 \quad x < 0$$

$$V(x) = V_0 \quad x > 0$$

- (a) Suppose the election has the energy $E > V_0$ and is incident from $-x$ direction, find the normalized wave function so it corresponds to unit incident flux.

- (b) Solve the above problem for the case $E < V_0$ and discuss the significance of the result with the help of a suitable example.

8. Write down the time independent Schrodinger equation for the $x < 0$ and $x > 0$ in case of a step potential

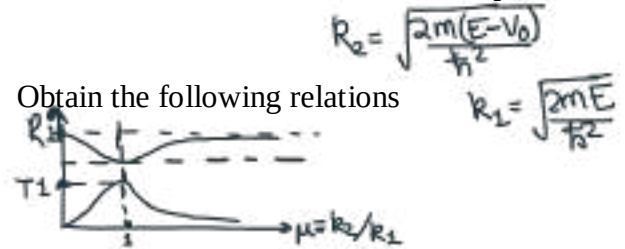
$$V(x) = 0 \quad x < 0$$

$$= V_0 \quad x \geq 0$$

Discuss the solutions thus obtained for the case $E > V_0$. Obtain the following relations

$$|R|^2 = \left(\frac{1-\mu}{1+\mu} \right)^2 \text{ and } \frac{K}{K_0} |T|^2 = \mu \left(\frac{2}{1+\mu} \right)^2$$

Where $\mu = \sqrt{1 - \frac{V_0}{E}}$, R and T are reflection and transmission coefficients respectively.



Other symbols have their usual meanings. Plot the behavior of $|R|^2$ and $\frac{K}{K_0} |T|^2$ with μ (1997)

9. An infinite square well potential is

$$V(x) = 0 \quad -\frac{a}{2} < x < \frac{a}{2}$$

$$= \infty \quad \frac{a}{2} < x < -\frac{a}{2}$$

Handwritten notes for problem 9:

$$\psi(x) = \frac{1}{\sqrt{a}} \cos(kx) \text{ or } \sin(kx) \Rightarrow \cos(k \frac{a}{2}) = 0 \Rightarrow k_n = \frac{(2n-1)\pi}{a}$$

$$\sin(k \frac{a}{2}) = 0 \Rightarrow k' = \frac{2m\pi}{a}$$

$$\Delta x = \sqrt{\langle x^2 \rangle - \langle x \rangle^2} = \frac{a}{\sqrt{12}}$$

$$\Delta p = \sqrt{\langle p^2 \rangle - \langle p \rangle^2} = \frac{h}{2a}$$

$$\Delta x \Delta p = \frac{a}{\sqrt{12}} \cdot \frac{h}{2a} = \frac{h}{2\sqrt{12}} \approx \frac{h}{4}$$

Solve the time independent Schrodinger's equation and find the closed form expressions for the eigen - values and eigen functions of such a potential. Give a schematic representation of the first three energy levels and corresponding wave functions and probability distribution functions. Show that the zero-point energy is in accord with the uncertainty principle. (1998)

10. For a particle confined in a one - dimensional potential well of length L the wave function is

$$\psi(x) = C \sin\left(\frac{\pi x}{L}\right) \quad 0 < x < L \quad \text{and } \psi(x) = 0, \text{ outside}$$

Calculate the expectation values of x and p.

defn expectation values
→ solve / make graph
→ assumption

(1999)

11. A 500 μA beam of electrons of kinetic energy 1.5eV enter a region with a sharply defined boundary in which their energy is reduced to 1.0eV by a difference of potential. Determine the reflected and transmitted currents. Derive the formulae used. (2000)

12. Show that the motion of a classical particle is analogous to the motion of a wave packet, which can be constructed by superposition of a large number of plane waves. Construct such wave packet and derive an expression for its group velocity. (2000)

13. A particle is in a quantum system with a parabolic potential well.

(i) Using appropriate method find the ground state energy.

(ii) Obtain an expression for the ground state.

Handwritten notes for problem 13:

$$\psi_0 = \left(\frac{m\omega}{\pi\hbar} \right)^{1/4} \exp\left(-\frac{m\omega x^2}{2\hbar}\right)$$

$$\hat{A} \psi_0 = \sum a_k \hat{A} \phi_k = |g\rangle = \sum a'_k \phi_k$$

$$|g\rangle = \sum a'_k \phi_k$$

$$a'_m = \sum a_k \langle \phi_m | \hat{A} | \phi_k \rangle$$

(2001)

14. (i) Write down the matrix representation of the energy operator of a linear oscillator.

(ii) A linear oscillator is prepared in the state given by:

$$\psi(x) = 1/\sqrt{5} \{ \psi_0(x) + \sqrt{2} \psi_1(x) + \sqrt{2} \psi_2(x) \}$$

Evaluate the energy of the oscillator in this state.

Handwritten notes for problem 14:

$$\text{energy} = \langle \hat{H} \rangle = \frac{1}{5} [E_0 + 2E_1 + 2E_2] = 1.7 \hbar \omega$$

(since $\psi(x)$ is normalized)

$$\hat{H} = \begin{bmatrix} E_0 & 0 & 0 \\ 0 & E_1 & 0 \\ 0 & 0 & E_2 \end{bmatrix}$$

Handwritten notes for problem 14 (continued):

$$\hat{A} \psi_0 = \sum a_k \hat{A} \phi_k = |g\rangle = \sum a'_k \phi_k$$

$$|g\rangle = \sum a'_k \phi_k$$

$$a'_m = \sum a_k \langle \phi_m | \hat{A} | \phi_k \rangle$$

in some basis

in eigenbasis

matrix representation

15. The wave function of a particle is

$$\psi(x) = A \exp(-x^2/a^2 + ikx).$$

Find the expectation values of position (x) and momentum (p) for the particle.

(2002)

16. Solve the one dimensional Schrodinger wave equation with potential :

$$V(x) = \begin{cases} 0 & \text{for } x < -a \\ V & \text{for } -a < x < a \\ 0 & \text{for } x > a \end{cases}$$

(2002)

17. Distinguish between a classical and a quantum mechanical harmonic oscillator. Explain the existence of zero point energy.

(2002)

18. A particle of mass m , with energy E such that $-V_0 < E < 0$, is trapped in a potential well as shown below:

(i) Write time independent Schrodinger equation in regions

(a) $0 < x < a$ and (b) $x > a$

(ii) Obtain an expression from which the energy eigen values can be determined.

(iii) Show that for at least one bound state to exist $a^2 V_0 \geq \hbar^2 \pi^2 / 8m$.

(2003)

19. Consider a particle in a one – dimensional box of length a in a quantum state at $t = 0$, given by

$$\Psi(q, 0) = \begin{cases} A; 0 < q < a \\ 0; \text{otherwise} \end{cases} \quad \text{Where } A = \text{a real constant. Write the expression for } \psi(q, t) \text{ Obtain the value of } A \text{ and expansion coefficient of the total wave function.}$$

(2004)

20. The ground state wave function of a linear simple harmonic oscillator is $\psi = A e^{-\alpha x^2/2}$

Calculate the constant A and the average values of x^2 and x . Given that $\int_0^\infty e^{-x^2} = \frac{\pi^{1/2}}{2}$

(2005)

21. Consider a particle of mass m in an infinite one dimensional potential well of width a . the particle is found in the state given by $\psi(x) = c \left[\sin \frac{\pi x}{a} + \frac{1}{2} \sin \frac{2\pi x}{a} \right]$ Calculate c . If a measurement of energy is made, what are the possible results and what are the probabilities for each one of them?

(2005)

22. (a) Distinguish between potential well and potential barrier. Give their illustrations. Considering one dimensional potential step, prove that sum of the reflection and transmission coefficients is unity.

(b) Prove that $\frac{d}{dt} \langle x \rangle = \frac{1}{m} \langle p_x \rangle$

Define all the terms of this relation and give its physical interpretation.

(2006)

23. Solve the Schrodinger equation for a linear harmonic oscillator; obtain the eigenvalues and the corresponding eigen functions.

(2007)

24. A one – dimensional potential barrier is represented by the function $V(x) = 0$ for $x < 0 = V_0$ for $x > 0$ where V_0 is positive. Find the transmission coefficient for particles of mass m incident from the left on the barrier. (Same as below) (2007)

25. Show that the probability of transmission across the step barrier represented by the potential $V(x) = \begin{cases} 0 & \text{for } x < 0 \\ V_0 & \text{for } x > 0 \end{cases}$ is $T = \frac{4k_1 k_2}{(k_1 + k_2)^2}$ where k_1 and k_2 are wave numbers in regions $x < 0$ and $x > 0$. (2008)

26. An electron is moving freely in a one dimensional infinite potential box with walls at $x = 0$ and $x = a$. If the electron is initially in the ground state of the box and if suddenly the wall at $x = a$ is moved $x = 4a$, calculate the probability of finding the particle in the ground state of the new box. (2008)

27. The Hamiltonian of a particle moving along the x -axis is given by $\hat{H} = -\alpha \frac{d^2}{dx^2} + 16\alpha x^2$, Where α is a real and positive constant having dimensions of energy. (2008)

- If $\psi(x) = Ae^{-2x^2}$, find the normalization constant A . Check whether ψ is an eigen function of $\hat{H}$. If yes, find the corresponding eigen value.
- Calculate the probability of finding the particle anywhere along the negative x -axis.
- Find the eigen value of $\hat{H}$ corresponding to the eigen function $\phi(x) = x\psi(x)$, where $\psi(x)$ is the same as in part (i).
- Are the wave functions $\psi(x)$ and $\phi(x)$ orthogonal?

28. Show that the time-dependent part of all the solutions of the Schrodinger equation in one-dimension has the structure $\phi(t) = \exp(-iEt/\hbar)$, provided the potential is not an explicit function of time. (2009)

29. Consider a positron in a box. If the energy released is 60 eV when it jumps from the third excited state to the ground state, show that the width of the potential is nearly 0.3 nm. (2009)

30. Obtain an expression for the probability current for the plane wave $\psi(x, t) = \exp[i(kx - \omega t)]$. Interpret your result. (2010)

31. A system is described by the Hamiltonian operator, $H = -\frac{d^2}{dx^2} + x^2$ Show that the function $A x \exp\left(-\frac{x^2}{2}\right)$ is an eigen function of H . Determine the eigen values of H . (2011)

32. Solve the Schrodinger equation for a particle of mass m in an infinite rectangular well defined by $V(x) = 0$ $0 \leq x \leq L = \infty$ $x < 0, x > L$ Obtain the normalized eigen functions and the corresponding eigen values. (2011)

33. Normalize the wave function $\psi(x) = e^{-|x|} \sin \alpha x$ (2011)

34. (a)

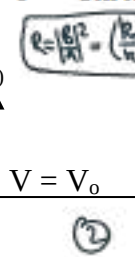
$$\psi_1(x) = A e^{ik_1 x} + B e^{-ik_1 x}$$

$$\psi_2(x) = C e^{ik_2 x} + D e^{-ik_2 x}$$

$D=0$ for no reflection by transmitted particle

$$\Rightarrow (A+B)C = (A-B)k_2 = k_2 C$$

$V(x)$

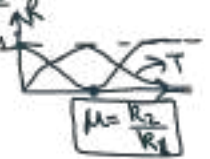


$$R = \frac{|B|^2}{|A|^2} = \left(\frac{k_2 - k_1}{k_2 + k_1} \right)^2$$

$$T = \frac{4k_1 k_2}{(k_1 + k_2)^2}$$

$$k_1 = \sqrt{2m(E - V_0)} \quad \& \quad k_2 = \sqrt{2mE}$$

$$E \rightarrow V_0, R=1 \quad \& \quad R=0 (E \rightarrow \infty)$$



Consider a beam of particles incident on one-dimensional step function potential with energy $E > V_0$ as shown in the above figure. Solve the Schrodinger equation and obtain expressions for the reflection and transmission coefficients.

(b) What are the limits of the reflection coefficient for $E \rightarrow V_0$ and $E \rightarrow \infty$?

(c) Discuss the cases $0 < E < V_0$ and $E < 0$.

(2013)

35. Obtain the time independent Schrödinger equation for a particle. Hence deduce the time independent Schrödinger equation.

$$-\frac{\hbar^2}{2m} \nabla^2 \psi + V\psi = E\psi$$

36. Solve the Schrödinger equation for a particle of mass m confined in one dimensional potential well of the form: $V=0$, when $0 \leq x \leq L$, infinity, when $x < 0$, $x > L$. Obtain the discrete energy values and the normalized eigen functions. (2014)

37. An electron is moving in a one dimensional box of infinite height and width 1 Å (Angstrom). Find the minimum energy of electron. (2014)

38. Normalized wave function of a particle is given: $\psi(x) = A \exp(-x^2/a^2 + ikx)$. Find the expectation value of position. (repeated) (2015)

39. Solve the Schrodinger equation for a step potential and calculate the transmission and reflection coefficient for the case when the kinetic energy of the particle E_0 is greater than the potential energy V (i.e., $E_0 > V$). (2016)

40. Calculate the lowest energy of an electron confined to move in a 1-dimensional potential well of width 10nm. (2016)

41. Using Schrödinger equation, obtain the eigenfunctions and eigenvalues of energy for a 1-dimensional harmonic oscillator. Sketch the profiles of eigenfunctions for first three energy states. (2017)

42. Calculate the probability of transmission of an electron of 1.0 eV energy through a potential barrier of 4.0 eV and 0.1 nm width. (2017)

43. The wave function of a particle is given as $\psi(x) = \frac{1}{\sqrt{a}} e^{-|x|/a}$. Find the probability of locating the particle in the range $-a \leq x \leq a$. (2018)

44. Calculate the zero-point energy of a system consisting of a mass of 10^{-3} kg connected to a fixed point by a spring which is stretched by 10^{-2} m by a force of 10^{-1} N . The system is constrained to move only in one direction. (2018)

$L = 10^{-2} \text{ m}$
 $F = 10^{-1} \text{ N}$
 $F = KL$
 $K = \frac{F}{L} = \frac{10^{-1}}{10^{-2}} = 10 \text{ N/m}$
 $K = m\omega^2$
 $\omega^2 = \frac{F}{mL} \Rightarrow E_0 = \frac{\hbar\omega}{2}$

45. The general wave function of harmonic oscillator (one-dimensional) are of the form

$$u_n(x) = \sum_{k=0}^{\infty} a_k y^k e^{-y^2/2}$$

With $y = \sqrt{\frac{m\omega}{\hbar}} x$, and coefficients a_k are determined by recurrence relations

$$a_{k+2} = \frac{2(k-n)}{(k+1)(k+2)} a_k$$

Corresponding energy levels are $E_n = \left(n + \frac{1}{2}\right) \hbar\omega$. Discuss the parity of these wave functions. What happens, if the potential for $x \leq 0$ is infinite (half harmonic oscillator)? (2018)

$a_{k+2} = \frac{2(k+1 - \frac{2E}{\hbar\omega})}{(k+1)(k+2)}$
 $\frac{2E}{\hbar\omega} = 2n+1$
for $k=n$,
odd for $x \leq 0, \psi \rightarrow \infty$ means $\psi(x=0)=0$
even

46. Which of the following functions is/are acceptable solution(s) of the *Schrödinger equation*? (2018)

(i) $\psi(x) = A e^{-ikx} + B e^{ikx}$

(ii) $\psi(x) = A e^{-kx} + B e^{kx}$

(iii) $\psi(x) = A \sin 3kx + B \cos 5kx$

(iv) $\psi(x) = A \sin 3kx + B \sin 5kx$

(v) $\psi(x) = A \tan kx$

Explain your answer.

Born approximation,
 ψ should be continuous, square integrable

only odd $n \rightarrow$ none of even ψ is finite eigenfn in zero at $x=0$

$\psi(x) = A \left(\frac{e^{i3kx} - e^{-i3kx}}{2i} \right) + B \left(\frac{e^{i5kx} + e^{-i5kx}}{2} \right)$

47. A beam of particles of energy 9 eV is incident on a potential step 8 eV high from the left. What percentage of particles will reflect back? (2018)

$R = \left[\frac{R_1 - R_2}{R_1 + R_2} \right]^2 = \frac{1}{289}$

48. Write down the Hamiltonian operator for a linear harmonic oscillator. Show that the energy eigenvalue of the same can be given by $E_n = \left(n + \frac{1}{2}\right) \hbar\omega_0$ at energy state n with ω_0 being the natural frequency of vibration of the linear oscillator. Prove that $n = 0$ energy state has a wave function of typical Gaussian form. (2019)

$H = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} + V$

$\psi = \sqrt{\frac{m\omega}{\pi\hbar}} e^{-\frac{m\omega x^2}{2\hbar}}$
 $\frac{\partial^2 \psi}{\partial x^2} + \left(\frac{2E}{\hbar\omega} - 1\right)\psi = 0 \rightarrow \frac{\partial^2 \psi}{\partial x^2} + \left(\frac{2E}{\hbar\omega} - 1\right)\psi = 0$
 $\frac{\partial^2 \psi}{\partial x^2} - 2\psi \frac{\partial^2}{\partial x^2} + \left(\frac{2E}{\hbar\omega} - 1\right)\psi = 0$
 $\frac{\partial^2 \psi}{\partial x^2} - 2\psi \frac{\partial^2}{\partial x^2} + \left(\frac{2E}{\hbar\omega} - 1\right)\psi = 0$
 $\frac{\partial^2 \psi}{\partial x^2} - 2\psi \frac{\partial^2}{\partial x^2} + \left(\frac{2E}{\hbar\omega} - 1\right)\psi = 0$
 $\frac{\partial^2 \psi}{\partial x^2} - 2\psi \frac{\partial^2}{\partial x^2} + \left(\frac{2E}{\hbar\omega} - 1\right)\psi = 0$

49. Estimate the size of hydrogen atom and the ground state energy from the uncertainty principle. (2019)

$\Delta x \Delta p \rightarrow$ define $\rightarrow$ write & solve $\rightarrow$ significance

Don't use

uncertainty means spreading position & momenta in a way, $\Delta E \neq 0$

$A^3 = I$
 $A|\psi\rangle = \alpha|\psi\rangle \Rightarrow A^3|\psi\rangle = \alpha^3|\psi\rangle \Rightarrow \alpha^3 = 1$
 $\alpha = 1, e^{i\frac{2\pi}{3}}, e^{-i\frac{2\pi}{3}}$
 $\alpha = 1, e^{i\frac{2\pi}{3}}, e^{-i\frac{2\pi}{3}}$
 $\alpha = 1, e^{i\frac{2\pi}{3}}, e^{-i\frac{2\pi}{3}}$

$|\psi\rangle = \begin{bmatrix} 1 \\ 0 \end{bmatrix}$

If the z -component of an electron spin is $+\frac{\hbar}{2}$, what is the probability that its component along a direction z' (forming an angle θ with z -axis) is $\frac{\hbar}{2}$ or $-\frac{\hbar}{2}$?
What is the average value of spin along z' ?

$$S_x = S_y = S_z = \frac{\hbar}{2} \begin{pmatrix} \cos \theta & \sin \theta \\ -\sin \theta & \cos \theta \end{pmatrix} + \frac{\hbar}{2} \begin{pmatrix} \cos \theta & \sin \theta \\ -\sin \theta & \cos \theta \end{pmatrix} + \frac{\hbar}{2} \begin{pmatrix} \cos \theta & \sin \theta \\ -\sin \theta & \cos \theta \end{pmatrix}$$

$$\left(\begin{pmatrix} \cos \theta & \sin \theta \\ -\sin \theta & \cos \theta \end{pmatrix} - \begin{pmatrix} \cos \theta & \sin \theta \\ -\sin \theta & \cos \theta \end{pmatrix} \right) = \begin{pmatrix} 0 & 0 \\ 0 & 0 \end{pmatrix} = \begin{pmatrix} 0 & 0 \\ 0 & 0 \end{pmatrix}$$

$$S^2 = \frac{\hbar^2}{4} (\cos^2 \theta + \sin^2 \theta) = \frac{\hbar^2}{4} (1) = \frac{\hbar^2}{4}$$

$$\frac{\hbar}{2} \begin{pmatrix} \cos \theta & \sin \theta \\ -\sin \theta & \cos \theta \end{pmatrix} \begin{pmatrix} 1 \\ 0 \end{pmatrix} = \frac{\hbar}{2} \begin{pmatrix} \cos \theta \\ -\sin \theta \end{pmatrix} = \frac{\hbar}{2} \begin{pmatrix} \cos \theta \\ -\sin \theta \end{pmatrix}$$

Upper sign: Use $1 - \cos \theta = 2 \sin^2 \frac{\theta}{2}$, $\sin \theta = 2 \sin \frac{\theta}{2} \cos \frac{\theta}{2}$. Then $S = e^{-i\phi} \frac{\hbar}{2} \begin{pmatrix} \cos \theta & \sin \theta \\ -\sin \theta & \cos \theta \end{pmatrix} e^{i\phi}$. Normalizing

$$1 = |\alpha|^2 + |\beta|^2 = |\alpha|^2 + \frac{\sin^2(\theta/2)}{\cos^2(\theta/2)} |\alpha|^2 = |\alpha|^2 \frac{1 + \tan^2(\theta/2)}{\cos^2(\theta/2)} = |\alpha|^2 \frac{1}{\cos^2(\theta/2)}$$

Lower sign: Use $1 + \cos \theta = 2 \cos^2 \frac{\theta}{2}$, $\sin \theta = 2 \sin \frac{\theta}{2} \cos \frac{\theta}{2}$. Then $S = e^{-i\phi} \frac{\hbar}{2} \begin{pmatrix} \cos \theta & \sin \theta \\ -\sin \theta & \cos \theta \end{pmatrix} e^{i\phi}$. Normalizing

$$1 = |\alpha|^2 + |\beta|^2 = |\alpha|^2 + \frac{\sin^2(\theta/2)}{\cos^2(\theta/2)} |\alpha|^2 = |\alpha|^2 \frac{1 + \tan^2(\theta/2)}{\cos^2(\theta/2)} = |\alpha|^2 \frac{1}{\cos^2(\theta/2)}$$

Problem 4.31

$$\text{Then we have states } \chi_+ = \begin{pmatrix} 1 \\ 0 \end{pmatrix}, \chi_- = \begin{pmatrix} 0 \\ 1 \end{pmatrix}, \chi_0 = \begin{pmatrix} 1 \\ 0 \end{pmatrix}$$

Problem 4.31 Construct the matrix S_x representing the component of spin angular momentum along an arbitrary direction $\hat{r}$. Use spherical coordinates, so that

$$\hat{r} = \sin \theta \cos \phi \hat{i} + \sin \theta \sin \phi \hat{j} + \cos \theta \hat{k} \quad (4.154)$$

Find the eigenvalues and (normalized) eigenspinors of S_x . Answer:

$$\chi_+^{(x)} = \begin{pmatrix} \cos(\theta/2) \\ e^{i\phi} \sin(\theta/2) \end{pmatrix}, \chi_-^{(x)} = \begin{pmatrix} -\sin(\theta/2) \\ e^{i\phi} \cos(\theta/2) \end{pmatrix} \quad (4.155)$$

Problem 4.32 Construct the spin matrices (S_x , S_y , and S_z) for a particle of spin $1/2$. Hint: How many eigenstates of S_z are there? Determine the action of S_x , S_y , and S_z on each of these states. Follow the procedure used in the text for spin $1/2$.

4.4.2 Electron in a Magnetic Field

A spinning charged particle constitutes a magnetic dipole. Its magnetic dipole moment μ is proportional to its spin angular momentum S :

$$\mu = \gamma S \quad (4.156)$$

the proportionality constant γ is called the gyromagnetic ratio.¹⁷ When a magnetic dipole is placed in a magnetic field B , it experiences a torque, $\mu \times B$, which tends to line it up parallel to the field (just like a compass needle). The energy associated with this torque is¹⁸

$$H = -\mu \cdot B \quad (4.157)$$

so the Hamiltonian of a spinning charged particle, in vac^2 is a magnetic field B , becomes

$$H = -\gamma B \cdot S \quad (4.158)$$

where S is the appropriate spin matrix (Equation 4.147, in the case of spin $1/2$).

$$S_{\alpha} = \hat{S} \cdot \hat{r} = (S_x \hat{i} + S_y \hat{j} + S_z \hat{k}) \cdot (\sin \theta \cos \phi \hat{i} + \sin \theta \sin \phi \hat{j} + \cos \theta \hat{k})$$

$$\text{or, } S_{\alpha} = \frac{\hbar}{2} \begin{pmatrix} \cos \theta & e^{-i\phi} \sin \theta \\ e^{i\phi} \sin \theta & -\cos \theta \end{pmatrix}$$

$$\text{eigenfn} \rightarrow \frac{\hbar}{2} \begin{pmatrix} \cos \theta & e^{-i\phi} \sin \theta \\ e^{i\phi} \sin \theta & -\cos \theta \end{pmatrix} \begin{pmatrix} \alpha \\ \beta \end{pmatrix} = \pm \frac{\hbar}{2} \begin{pmatrix} \alpha \\ \beta \end{pmatrix}$$

$$\text{For } \lambda = \frac{\hbar}{2}; \quad \beta = \alpha \frac{e^{i\phi} \sin(\theta/2)}{\cos(\theta/2)} \quad \& \alpha^2 + \beta^2 = 1$$

$$\hookrightarrow \alpha = \cos(\theta/2), \quad \beta = e^{i\phi} \sin(\theta/2)$$

$$\chi_+ = \begin{pmatrix} \cos(\theta/2) \\ e^{i\phi} \sin(\theta/2) \end{pmatrix}$$

$$|\chi\rangle = \begin{bmatrix} 1 \\ 0 \end{bmatrix}$$

$$\text{for } \lambda = -\frac{\hbar}{2}; \quad \chi_- = \begin{pmatrix} e^{-i\phi} \sin(\theta/2) \\ -\cos(\theta/2) \end{pmatrix}$$

$$P_{-\frac{\hbar}{2}} = |\langle \chi_- | \chi \rangle|^2 = \left| \begin{bmatrix} e^{-i\phi} \sin(\theta/2) & -\cos(\theta/2) \end{bmatrix} \begin{bmatrix} 1 \\ 0 \end{bmatrix} \right|^2 = \sin^2(\theta/2) \quad \text{Ans}$$

$$P_{+\frac{\hbar}{2}} = |\langle \chi_+ | \chi \rangle|^2 = \cos^2(\theta/2) \quad \text{Ans}$$

$$\langle S_{\alpha} \rangle = \frac{\hbar}{2} \begin{bmatrix} 1 & 0 \end{bmatrix} \chi_+ \begin{bmatrix} \cos \theta & e^{-i\phi} \sin \theta \\ e^{i\phi} \sin \theta & -\cos \theta \end{bmatrix} \begin{bmatrix} 1 \\ 0 \end{bmatrix} = \frac{\hbar}{2} \cos \theta \quad \text{Ans}$$

TUTORIAL SHEET: 3 Quantum Mechanics-II

1. (i) Find the eigen states of angular moment vector's component L_z of a spherically symmetric system. (ii) Write down any two properties of Pauli spin matrices, after defining them through suitable expression. (2001)

2. Write down the eigen value and spin state of an electron for the spin operator S . (i) Show that the spin states of electron are orthogonal to each other. (ii) State the spin angular momentum commutation relations (iii) Give the explicit forms of all the Pauli spin matrices. (2001)

3. (i) Express the Cartesian components of the angular momentum L in operator form. Show that $[L^2, L_z]=0$. What is the significance of this commutation relation? (ii) Show that the Pauli matrices anti-commute. (2003)

4. Show that the radial probability density for the ground state of the hydrogen atom has a maximum at $r = a$. The ground state wave function of the hydrogen atom is given by $\Psi(r) = 1/\sqrt{\pi} a^{3/2} e^{-r/a}$, where the Bohr radius. (2003)

5. N number of non-interacting electrons are confined in a cube of volume L^3 . Obtain an expression for the Fermi energy. (2003)

6. A hydrogen atom is in the following state $\Psi_{nlm}(r,0) = \frac{1}{\sqrt{14}} [2\Psi_{100}(r) - 3\Psi_{200}(r) + \Psi_{322}(r)]$ (i) What is the probability of finding the system in the state 200? (ii) What are H and L_z ? (2004)

7. Show that for the ground state of hydrogen atom the mean value of r is $3/2 a_0$, where a_0 is Bohr radius. (2004)

8. Solve the eigen value equation $L^2 Y(\theta, \Phi) = \lambda \hbar^2 Y(\theta, \Phi)$ and obtain the eigen value of L^2 . (2004)

9. Write the commutation relations for the position variable x and the momentum component p_x, p_y and p_z . Explain the physical significance of these relations. (2005)

10. The wave function of a particle confined in a cube of volume L^3 is given by $\psi(x, y, z) = \left(\frac{2}{L}\right)^{3/2} \sin \frac{\pi x}{L} \sin \frac{\pi y}{L} \sin \frac{\pi z}{L}$ Calculate the average values of p_x and p_x^2 in the region $0 < x < L$. (2005)

11. Set up the time-independent Schrodinger equation for an electron moving in a Coulomb field, $V(r) = -\frac{Ze^2}{4\pi\epsilon_0 r}$, in polar coordinates. Solve the radial equation to get the energy eigen values. (2005)

$$p = \frac{Z e^2}{a n}, \quad a_0 = \frac{4\pi\epsilon_0 \hbar^2}{m e^2}, \quad r_n = a_0 n^2$$

11. Explain how the problem of the hydrogen atom could be solved using Schrodinger equation. Also derive an expression for its energy eigenvalue and discuss the associated bound states of this case. (2006)

12. (i) Define angular momentum. Express it in operator form and show that $\vec{L} \times \vec{L} = i\hbar \vec{L}$ Explain the physical significance of this relation. $\rightarrow [L_x, L_y] = i\hbar L_z$ as L_z is not zero as components of $\vec{L}$ don't commute i.e. not measurable simultaneously (2007)

- (ii) Let Y_{lm} be an eigenstate of L^2 and L_z with eigenvalues $l(l+1)\hbar^2$ and $m\hbar$, respectively. Show that $\phi = (L_x + iL_y)Y_{lm}$ is likewise an eigenstate of L^2 and L_z and determine the eigenvalues. $[L^2, L_{\pm}] = [L^2, L_x \pm iL_y] = 0$ & $[L_z, L_{\pm}] = \pm \hbar L_{\pm}$ $\rightarrow [L_z, L_x + iL_y] = \hbar(L_x - iL_y) = \hbar \phi$ $\rightarrow L_z \phi = (l+1)\hbar \phi$ $\rightarrow \phi$ is an eigenstate of L_z with eigenvalue $(l+1)\hbar$ $L^2 \phi = l(l+1)\hbar^2 \phi$ $\rightarrow \phi$ is an eigenstate of L^2 with eigenvalue $l(l+1)\hbar^2$

13. Determine the discrete energy levels and the corresponding eigenfunctions for a particle in an infinitely deep potential well inside a cube of dimension L , i.e. assume $V(x, y, z) = 0$ for $0 < x < L$; $0 < y < L$; $0 < z < L = \infty$ elsewhere (Repetitive) (2007)

14. In the free electron theory of metals, a conductor is regarded as consisting of free electrons in a three dimensional box. Using the results of Q(14), obtain an expression for the density of states. (2007)

15. An electron is in the spin state $\chi = A \begin{pmatrix} 3i \\ 4 \end{pmatrix}$ Determine the normalization constant A . Find the expectation value of the spin operator $\hat{S}_x$ and also the uncertainty in the value of S_x in this state. $\rightarrow \chi^\dagger \chi = A^\dagger A \begin{pmatrix} 3i \\ 4 \end{pmatrix} \begin{pmatrix} 3i \\ 4 \end{pmatrix} = 1 \Rightarrow A = \frac{1}{5}$ $\langle S_x \rangle = \chi^\dagger \hat{S}_x \chi = \frac{1}{2} \chi^\dagger \sigma_x \chi = 0$ $\langle S_x^2 \rangle = \frac{\hbar^2}{4}$, $\Delta S_x = \sqrt{\langle S_x^2 \rangle - \langle S_x \rangle^2} = \frac{\hbar}{2}$ (2008)

16. Write (do not derive) the formula for the energy levels of a particle in a three dimensional cubical box of side L . How many electrons can occupy the level having energy $66\hbar^2/8mL^2$? (2008)

$E_n = \frac{\hbar^2}{8mL^2} (n_x^2 + n_y^2 + n_z^2) = \frac{66\hbar^2}{8mL^2} \Rightarrow (n_x, n_y, n_z) = \left(\frac{66}{3}, \frac{66}{3}, \frac{66}{3} \right) = (22, 22, 22)$ $\rightarrow 12$ states $\rightarrow 24 e^-$

17. Write the commutation relations of angular momentum operator $\hat{L}_x, \hat{L}_y$ and $\hat{L}_z$ and calculate the

commutators $[\hat{L}_+, \hat{L}_z]$ and $[\hat{L}_-, \hat{L}_z]$. Show that $\hat{L}_+ |l, m\rangle = \sqrt{l(l+1) - m(m+1)} |l, m+1\rangle$, same as (17)(ii)

Where $|l, m\rangle$ is the state with definite values for L^2 and L_z . (2008)

18. (i) The quantum mechanical probability distribution function of an electron in the ground state of the hydrogen atom is $P(r) = Nr^2 \exp(-2br)$. Using the result $\int_0^\infty P(r) dr = 1$, deduce that N is

proportional to b^3 .

- (ii) Prove that the value of $40 k_B T$ at $T = 300$ K is nearly 1 eV. Hence determine the Fermi temperature of metal whose Fermi energy is 9.4 eV. $\rightarrow \text{explain state associated} \Rightarrow \int P(x) dx = 1$ & $\boxed{x = 2b\pi} \Rightarrow N \propto b^3$ $K_B T_F = 9.4 \text{ eV}$ & $\frac{1}{2} m v_F^2 = E_F$

- (iii) Show that the Fermi velocity is related to the Fermi energy of electrons through the Relation

$\frac{v_F}{c} = 1.98 \left(\frac{E_F}{1 \text{ MeV}} \right)^{1/2}$ $\left\{ \frac{v_F}{c} = \sqrt{\frac{2E_F}{m c^2}} = \sqrt{\frac{2E_F}{0.51 \text{ MeV}}} = 1.98 \sqrt{\frac{E_F}{1 \text{ MeV}}} \right\}$ (2009)

19. Prove that the most probable distance of an electron from the proton (in the hydrogen atom) is the Bohr radius of the hydrogen atom. Consider only the ground state. (2009)

Start with ψ_{1s} & their graph $\psi_{1s} = \frac{1}{(\pi a_0^3)^{1/2}} \exp\left(-\frac{r}{a_0}\right)$

radial probability $= 4\pi r^2 dr \times |\psi|^2$ $\frac{dP}{dr} \Big|_{r=r_0} = 0$ & $\frac{d^2P}{dr^2} < 0 \Rightarrow \boxed{r_0 = a_0}$

$E = \frac{\hbar^2 k^2}{2m} (n_x^2 + n_y^2 + n_z^2)$ & assuming sphere with radius b/w n & $n+dn$
since n_x, n_y & n_z are +ve, $dN = \frac{4\pi n^2 dn}{8}$

20. (i) Consider a particle in a three-dimensional box. Derive an expression for $g(E)$, the density of states.

(ii) Show that $\frac{g(p)}{g(E)} = \frac{dE}{dp}$, where $g(p)$ is the density of states in the momentum space. Deduce that $g(p)$ is proportional to p^2 for a free non-relativistic particle.

$\{g(p)dp = g(E)dE\}$ (2009)

21. Show that the Pauli Spin Matrices obey the following relation:

i. $Tr(\sigma_x) = Tr(\sigma_y)$

ii. $\det(\sigma_y) = \det(\sigma_z)$

iii. The eigenvalues of σ_z and σ_x are the same.

iv. Write down the y-component of the spin angular momentum matrix corresponding to an antineutrino.

(2009)

22. Show that the Pauli spin matrices satisfy the following:

$\sigma_x^2 = \sigma_y^2 = \sigma_z^2 = 1$

$\sigma_x \sigma_y = -\sigma_y \sigma_x = i\sigma_z$

$\sigma_y \sigma_z = -\sigma_z \sigma_y = i\sigma_x$

$\sigma_z \sigma_x = -\sigma_x \sigma_z = i\sigma_y$

(2010)

23. The normalized wave function for the electron in hydrogen atom for the ground state is

$\psi(r) = \left(\frac{2}{a_0}\right)^{3/2} \exp\left(-\frac{r}{a_0}\right)$ Where a_0 is the radius of the first Bohr orbit. Show that the most

probable position of the electron is a_0

(2010)

24. Let $\vec{\sigma}$ be the vector operator with component equal to pauli spin matrices $\sigma_x, \sigma_y, \sigma_z$ if $\vec{a}, \vec{b}$ are vectors in 3D space, prove the identity $(\vec{\sigma} \cdot \vec{a})(\vec{\sigma} \cdot \vec{b}) = \vec{a} \cdot \vec{b} + i\vec{\sigma} \cdot (\vec{a} \times \vec{b})$.

(2011)

25. For hydrogen atom, ground state is

Calculate $\langle r \rangle$ and $\langle \frac{1}{r} \rangle$

(2011)

26. Show (i) $\sin(\sigma_x \theta) = \sigma_x \sin \theta$

(2011)

(ii) $\cos(\sigma_x \theta) = \cos \theta$

(2012)

27. What is the degree of degeneracy of the energy – eigen Values. What happened if the spin of the electron is taken into account?

(2012)

$\sigma_x = \begin{pmatrix} 0 & 1 \\ 1 & 0 \end{pmatrix}, \sigma_y = \begin{pmatrix} 0 & -i \\ i & 0 \end{pmatrix}, \sigma_z = \begin{pmatrix} 1 & 0 \\ 0 & -1 \end{pmatrix}$

28. Solve the radial part of the time independent Schrodinger Equation for a hydrogen atom. Obtain its expression for energy Eigen values. (2012)

29. Using the definition $L = r \times p$ of the orbital angular momentum operator, evaluate $[L(x), L(y)]$ (2013)

30. The normalized wave function for the electron for the electron in the ground state of the hydrogen atom is given by $\psi(r) = \frac{1}{\sqrt{\pi a_0^3}} e^{-r/a_0}$. Find $\langle r \rangle$ and $\langle r^2 \rangle$. (2013)

31. If x and p are the position and momentum operators, prove the commutation relation $[p^2, x] = -2i\hbar p$ (2014)

32. Write down Pauli spin matrices. Express $J(x)$, $J(y)$ and $J(z)$ in terms of Pauli spin matrices. (2014)

33. Using the commutation relations

$$[x, p(x)] = [y, p(y)] = [z, p(z)] = i\hbar$$

Deduce the commutation relation between the components of angular momentum operator L .

$$[L(x), L(y)] = i\hbar L(z)$$

$$[L(y), L(z)] = i\hbar L(x) \text{ and}$$

$$[L(z), L(x)] = i\hbar L(y)$$

34. Solve the Schrodinger equation for a particle in a three dimensional rectangular potential barrier. (2014)

Explain the terms degenerate and non-degenerate states in this context.

35. A particle trapped in an infinitely deep square well of width a has a wave function

$$\psi = \left(\frac{2}{\pi}\right)^{\frac{1}{2}} \sin\left(\frac{\pi x}{a}\right)$$

The walls are suddenly separated by infinite distance. Find the probability of the

particle having momentum between p and $p + dp$

36. Write down the matrix representation of the three Pauli matrices σ_x , σ_y and σ_z . Prove that these matrices satisfy the following identities:

i. $[\sigma_x, \sigma_y] = 2i\sigma_z$

ii. $[\sigma^2, \sigma_x] = 0$

iii. $(\sigma \cdot A)(\sigma \cdot B) = A \cdot B + i\sigma \cdot (A \times B)$

If A and B commute with Pauli matrices.

(2016)

37. Calculate the density of states for an electron moving freely inside a metal with the help of quantum mechanical Schrodinger's equation for free particle in a box. (2016)

38. Evaluate the most probable distance of the electron from nucleus of a hydrogen atom in its $2p$ state. What is the probability of finding the electron at this distance? (2017)

$$P(r, \theta, \phi) d\tau = 4\pi r^2 dr |\psi|^2 = A e^{-2r/a_0} r^4 dr$$

$$\frac{dP}{dr} = \frac{d}{dr} (A e^{-2r/a_0} r^4) = 0 \Rightarrow r_{mp} = 4a_0$$

39. Explain why the square of the angular momentum (L^2) and only one of the components (L_x, L_y, L_z) of L are regarded as constants of motion. (2017)

40. The ground state wave function for hydrogen atom is

$$\psi(r) = \frac{1}{\sqrt{\pi a_0^3}} e^{-r/a_0}$$

Where a_0 is the Bohr radius. Sketch the wave function and the probability density as a function of the separation distance r . Calculate the probability that the electron in the ground state is found beyond the Bohr radius. (2018)

41. Prove the following identities:

(i) $[\hat{p}_x, \hat{L}_y] = i \hbar \hat{p}_z$
(ii) $e^{i(\hat{\sigma} \cdot \hat{n})\theta} = \cos\theta + i(\hat{\sigma} \cdot \hat{n})\sin\theta$

$$e^{i(\hat{\sigma} \cdot \hat{n})\theta} = \sum_{k=0}^{\infty} \frac{(i(\hat{\sigma} \cdot \hat{n})\theta)^k}{k!}$$

$[i(\hat{\sigma} \cdot \hat{n})\sin\theta] \rightarrow \text{odd + even terms}$
 $\rightarrow \cos\theta$

42. Show that for free electron gas, the density of states in three dimensions (3D) varies as $E^{1/2}$, and this dependence changes to E^0 for 2D (quantum well), $E^{-1/2}$ for 1D (quantum wire) and δ function for 0D (quantum dot). (2018)

43. How do you define density of states? Show that the density of states with wave vector less than $\vec{k}$ in a three-dimensional cubic box of volume V can be given by

$$D(\omega) = \frac{V}{2\pi^2} k^2 \left(\frac{dk}{d\omega} \right)$$

in the frequency spectrum between ω and $\omega + d\omega$. Here, assume that the number of modes per unit range of k is $L/(2\pi)$, L being the length of each side of the cubic box. (2019)

44. Define Pauli spin matrices σ_x , σ_y and σ_z . Using these definitions, prove the following:

(i) $\sigma_x^2 = \sigma_y^2 = \sigma_z^2 = 1$
(ii) $\sigma_x \sigma_y = i \sigma_z$; $\sigma_z \sigma_x = i \sigma_y$; $\sigma_y \sigma_z = i \sigma_x$

(2019)

45. Define angular momentum of a particle and find out the three components of the angular momentum operator $\hat{L}$ in Cartesian coordinates. Show that

$$\hat{L}^2 = -\hbar^2 \left[r^2 \nabla^2 - \frac{\partial}{\partial r} \left(r^2 \frac{\partial}{\partial r} \right) \right]$$

Prove that the operator $\hat{L}^2$ can also be expressed as

$$\hat{L}^2 = -\hbar^2 \left[\frac{1}{\sin\theta} \frac{\partial}{\partial \theta} \left(\sin\theta \frac{\partial}{\partial \theta} \right) + \frac{1}{\sin^2\theta} \frac{\partial^2}{\partial \phi^2} \right]$$

in spherical polar coordinates (r, θ, ϕ) .

(2019)

$$\textcircled{1} \vec{L} = -i\hbar \vec{r} \times \vec{\nabla} = -i\hbar \vec{r} \times \left(\hat{r} \frac{\partial}{\partial r} + \frac{\hat{\theta}}{r} \frac{\partial}{\partial \theta} + \frac{\hat{\phi}}{r \sin \theta} \frac{\partial}{\partial \phi} \right)$$

$$\vec{L} = -i\hbar \left(\hat{\phi} \frac{\partial}{\partial \theta} - \frac{\hat{\theta}}{\sin \theta} \frac{\partial}{\partial \phi} \right)$$

$$L^2 = -\hbar^2 \left(\hat{\phi} \frac{\partial}{\partial \theta} - \frac{\hat{\theta}}{\sin \theta} \frac{\partial}{\partial \phi} \right) \cdot \left(\hat{\phi} \frac{\partial}{\partial \theta} - \frac{\hat{\theta}}{\sin \theta} \frac{\partial}{\partial \phi} \right)$$

$$\left\{ \begin{aligned} \hat{r} &= \cos \theta \hat{z} + \sin \theta (\cos \phi \hat{x} + \sin \phi \hat{y}) \\ \hat{\theta} &= -\sin \theta \hat{z} + \cos \theta (\cos \phi \hat{x} + \sin \phi \hat{y}) \end{aligned} \right\} \quad \left\{ \begin{aligned} \hat{\phi} &= -\sin \phi \hat{x} + \cos \phi \hat{y} \end{aligned} \right\}$$

$$L^2 = -\hbar^2 \left[\frac{\partial^2}{\partial \theta^2} + \hat{\phi} \cdot \frac{\partial \hat{\phi}}{\partial \theta} \frac{\partial}{\partial \theta} - \hat{\phi} \cdot \frac{\partial \hat{\theta}}{\partial \theta} \times \frac{1}{\sin \theta} \frac{\partial}{\partial \phi} - \frac{\hat{\theta}}{\sin \theta} \cdot \frac{\partial \hat{\phi}}{\partial \phi} \frac{\partial}{\partial \theta} + \frac{1}{\sin^2 \theta} \frac{\partial^2}{\partial \phi^2} + \frac{\hat{\theta}}{\sin \theta} \cdot \frac{\partial \hat{\theta}}{\sin \theta} \frac{\partial}{\partial \phi} \times \frac{1}{\sin \theta} \frac{\partial}{\partial \phi} \right]$$

$$\left\{ \hat{\phi} \cdot \frac{\partial \hat{\phi}}{\partial \theta} = 0 = \hat{\phi} \cdot \frac{\partial \hat{\theta}}{\partial \theta} \quad \& \quad \hat{\theta} \cdot \frac{\partial \hat{\phi}}{\partial \phi} = -\cos \theta \quad \& \quad \hat{\theta} \cdot \frac{\partial \hat{\theta}}{\partial \phi} = 0 \right\}$$

$$\Rightarrow L^2 = -\hbar^2 \left[\frac{1}{\sin \theta} \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) + \frac{1}{\sin^2 \theta} \frac{\partial^2}{\partial \phi^2} \right] \rightarrow \text{required expression}$$

$$\nabla^2 = \frac{1}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial}{\partial r} \right) - \frac{1}{\hbar^2} L^2$$

Hydrogen Atom

$$\frac{-\hbar^2}{2\mu} \left[\frac{\partial^2}{\partial r^2} \left(r^2 \frac{\partial}{\partial r} \right) + \frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) + \frac{\partial^2}{\partial \phi^2} \right] \Phi = (E - V) R \Theta \Phi$$

$$\left\{ \frac{1}{\Phi} \frac{\partial^2 \Phi}{\partial \phi^2} = -m^2 \Rightarrow \Phi(\phi) = e^{\pm i m \phi} \right\} \left\{ \frac{-\hbar^2}{2\mu R} \frac{\partial}{\partial r} \left(r^2 \frac{\partial}{\partial r} \right) + (V - E) r^2 = \frac{\hbar^2}{2\mu r^2} \left[\frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) + \frac{\partial^2}{\partial \phi^2} \right] \right\}$$

for $R = \frac{U}{r} \Rightarrow \frac{\partial^2 U}{\partial r^2} + \frac{2\mu U}{\hbar^2} (E - V_{eff}) = 0$ where $V_{eff} = V + \frac{L(L+1)\hbar^2}{2\mu r^2}$

for $\rho = \sqrt{\frac{2\mu E}{\hbar^2}} r$ (as $E < 0$) $\Rightarrow \frac{d^2 U}{d\rho^2} = \left(1 - \frac{\rho_0}{\rho} + \frac{L(L+1)}{\rho^2} \right) U$

where $\frac{V}{E} = \frac{\rho_0}{\rho} = -\frac{Ze^2}{4\pi\epsilon_0 \hbar^2 E}$

for $\rho \rightarrow 0 \Rightarrow U = D \rho^{L+1}$, $\rho^{-L}(x)$ for $\rho \rightarrow \infty \Rightarrow \frac{d^2 U}{d\rho^2} = -U \Rightarrow U = C \exp(-\rho)$

Writing complete solution by peeling off $\Rightarrow U = A \rho^{L+1} e^{-\rho} V(\rho) \Rightarrow$ Using power series solution, $V(\rho) = \sum a_j \rho^j$

$R_{nl}(\rho) = \frac{\rho^{L+1} e^{-\rho}}{r} V_n(\rho) \rightarrow$ Laguerre polynomial

for $\frac{\hbar^2}{2\mu r^2} \left[\frac{\partial}{\partial \theta} \left(\sin \theta \frac{\partial}{\partial \theta} \right) + \frac{\partial^2}{\partial \phi^2} \right] \Theta = L(L+1) \Theta$ & $x = \cos \theta$

as $V(\rho)$ has to be finite;
 $[a_{l+1} = \frac{(2(L+1) - \rho_0)}{(L+1)(L+2)} a_l] \Rightarrow \rho_0 = 2L$

$(1-x^2) \frac{\partial^2 \Theta}{\partial x^2} - 2x \frac{\partial \Theta}{\partial x} + \left(L - \frac{m^2}{(1-x^2)} \right) \Theta = 0 \Rightarrow L = L(L+1)$ from $|m| \leq L$ & $\Theta(\theta) = P_L^m(\cos \theta)$

$\frac{d^2 U}{d\rho^2} = -U \Rightarrow U = C \exp(-\rho)$

ATOMIC PHYSICS TUTORIAL SHEET: 4

$\rightarrow$ j-j coupling is dominant after $Z=57$

$6s^2 \rightarrow s_1=0, s_2=0$
 $s_1=\frac{1}{2}, s_2=\frac{1}{2}$
 $j_1=\frac{1}{2}, j_2=\frac{1}{2}$
 $J=1, 0$

7th Questⁿ
If j-j coupling is used

1. What is Lamb shift? Explain it by illustrating through a suitable energy level diagram. Has Lamb shift been observed in any atom other than hydrogen? (2001)

Stern Gerlach (Pauli's concept) (anomalous Zeeman effect)

2. Give briefly an account on the important historical developments to establish the concept of electron spin. (2001)

$J = L + S$ (Russett Saunders L.S)
 $J = L + S$ (Russett Saunders J-J)
 $J = L + S$ (Russett Saunders J-J)

3. On the basis of vector atom model, discuss briefly the L-S and J-J coupling schemes. (2001)

Conclude with $\rightarrow$ explains Zeeman effect
coupling breaks under strong B (Paschen-Back)

4. Write down expressions for spin angular momentum, orbital angular momentum and total angular momentum under L-S coupling scheme. (2001)

5. Write the spectral symbol of the term with $S = 1/2, J = 5/2$ and $g = 6/7$ (2002)

$g = 1 + \frac{J(J+1) + S(S+1) - L(L+1)}{2J(J+1)}$
 $J = 3$

6. Derive an expression for the spin-orbit interaction energy in one electron system. Calculate the energy separation between the levels $^2P_{1/2}$ and $^2P_{3/2}$. (2002)

$\Delta E_{so} = -\vec{L} \cdot \vec{S} = \frac{1}{2} \hbar^2 \frac{Z}{a_0 n} \frac{1}{n} \frac{1}{n} \frac{1}{n}$
 $\Delta E_{so} = \frac{1}{2} \hbar^2 \frac{Z}{a_0 n} \frac{1}{n} \frac{1}{n} \frac{1}{n}$

7. Write the electronic configuration of mercury ($Z=80$). Obtain the spectral term for the normal and the first excited configuration on the atom. (2002)

$Z=80 \Rightarrow 6s^2 \rightarrow j=0, s=0, l=0 \Rightarrow 1s_0$
 $6s^2 6p^2 \rightarrow j=1, s=1/2, l=1 \Rightarrow 3P_1$

8. Calculate the Larmor frequency of a spin $1/2$ particle in a magnetic field B. (2003)

$\mu_B = \frac{e \hbar}{2m} (C+2S)$
 $\omega = \frac{2\pi \nu}{h} \mu_B B$

9. Define the gyromagnetic ratio and obtain an expression for the Lande 'g' factor. What is the importance of 'g' in spectroscopy? (2003)

Importance $\rightarrow$ defn, write symbols, select rule

10. What is Zeeman Effect? How can it be understood on a quantum mechanical basis? Obtain an expression for Zeeman shift. (2004)

$\Delta E = \mu_B B$
 $\Delta E = \mu_B B$

11. Calculate the Zeeman shift observed in the normal Zeeman Effect when a spectral line a wave length $5000 \text{ \AA}$ is subjected to the magnetic field of 0.4 wb/m^2 . (2004)

$\Delta E = \mu_B B$
 $\Delta E = \mu_B B$

12. What are the term symbols for atoms with the following S & L quantum numbers. $S=1/2$ and $L=3$? (2004)

$2S+1 L_J$
 $2 \times \frac{1}{2} + 1 = 2$
 $2 \times 3 + 1 = 7$

13. How many lines occur in the multiplet arising from $^2P_{3/2} \rightarrow ^2S_{1/2}$ and $^2P_{1/2} \rightarrow ^2S_{1/2}$ transitions of alkali metal atoms placed in a weak magnetic fields why? (2005)

$\Delta E = \mu_B B$
 $\Delta E = \mu_B B$

14. Describe Stern-Gerlach experiment and discuss its implications. (2005)

Implication $\rightarrow$ presence of spin, space quantization, confirms spin in vector model, fine spin to explain Zeeman

15. What is Lamb shift? Has it been observed in any atom other than hydrogen? (2007)

already in (6)

16. Describe Stern- Gerlach experiment. Explain how it demonstrates the discrete nature of the magnetic moment of an atom. (2007)

✓ 17. Explain normal and anomalous Zeeman Effect. Obtain expression for Zeeman splitting of an alkali metal spectral line, and illustrate with an example. *(done)* *(done in 32/33)* *(done)* (2008)

✓ 18. For transition to the ground state what is the longest wavelength that can be emitted by hydrogen *(done)* $\rightarrow n=2 \rightarrow n=1 \Rightarrow \lambda = \frac{1.216 \times 10^{-7} \text{ m}}{1 - 1/4} = 91.2 \text{ nm}$ (2008)

✓ 19. What was the aim of Stern – Gerlach experiment? Why were silver atoms chosen instead of electrons in the experiment? What was the outcome of the experiment? *(same as Q4)* (2009)

✓ 20. $2^2S_{1/2}$ level in H atom is 1058 MHz above the $2^2P_{1/2}$ level
(i) What is this known as? *Lamb shift violating Dirac's formula*
(ii) Express the above frequency in cm^{-1} . *$\bar{\nu} = \frac{\nu}{c} = 0.035 \text{ cm}^{-1}$*
(iii) Calculate the energy difference between the above two levels in eV. *$\Delta E = 4.4 \mu\text{eV}$* (2009)

✓ 21. (i) Establish that $hc = 1240 \text{ eV} \cdot \text{nm} = 1240 \text{ MeV} \cdot \text{fm}$
(ii) The energy levels of a hydrogen atom are given by $E_n = (-1/n^2) \cdot \text{Ryd}$. *$E_n = -13.6 \text{ eV} / n^2$*
Where $1 \text{ Ryd} = hcR$. Show that $R = 1.097 \times 10^7 \text{ m}^{-1}$ What exactly is R? (2009)

✓ 22. (i) Find the ground state total orbital (L) and total spin (S) quantum numbers for nitrogen. *$\rightarrow 2p^3$*
(ii) Find the L and S values of the two “first excited states” of helium atom. *$1s^2 2s^2 \rightarrow 1s 2s$*
(b). (i) Explain spin-orbit coupling of an atomic electron.
(ii) Show that the 2p state in the H atom splits up into two substates due to spin-orbit coupling.
(iii) Calculate the energy of separation in eV, resulting from the spin-orbit coupling when the magnetic field experienced by the electron is 0.4 T. (2009)

✓ 23. What is Zeeman Effect? How it can be understood on quantum mechanical basis? Obtain an expression for Zeeman splitting of atomic energy levels in a magnetic field B. Explain the magnetic splitting of sodium D-lines. *(done)* (2010)

✓ 24. What is spin-orbit interaction? Calculate the energy shift due to spin-orbit interaction term in H-like system. Discuss the significance of this shift in relation to the fine structure of hydrogen spectral lines *(done)* *$\rightarrow$ Lamb shift* (2010)

✓ 25. Discuss the fine structure of hydrogen atom spectrum. Draw the compound doublet spectrum arising as a result between 2p and 2d levels. *$\rightarrow$ draw diagram $\rightarrow$ spin-orbit $\rightarrow$ Lamb shift* *parameter of splitting characterised by fine structure constant $\alpha = \frac{e^2}{4\pi\epsilon_0\hbar c} = \frac{1}{137}$* (2011)

✓ 26. What do you mean by term symbols? Obtain term symbols for the following sets of values of S and L:
(i) $S = \frac{1}{2}, L = 2$ (ii) $S = 1, L = 1$ (iii) $S = \frac{3}{2}, L = 1$ *$\rightarrow$ Energy levels of e^- in atoms $\rightarrow$ mathematically $2S+1 L_J$* *$\rightarrow$ Succinctness & use in Bohr's scheme* (2011)

✓ 27. Show that $2S_{\frac{1}{2}}, 2P_{\frac{1}{2}}, 2P_{\frac{3}{2}}$ levels of sodium spectrum are split in the ratio of 3 : 1 : 2 due to anomalous Zeeman Effect. *$\rightarrow$ defⁿ it $\rightarrow$ write formula $\rightarrow$ ratio split calculation $\rightarrow$ conclude $\rightarrow$ weak B* *$\rightarrow$ g values $[g_{2P_{3/2}} = 4/3, g_{2P_{1/2}} = 2/3, g_{2S_{1/2}} = 2]$* *$\rightarrow$ g comes due to precession of angular momentum under B.* (2011)

28. Sodium doublets are produced by transitions $3^2P_{\frac{1}{2}} \rightarrow 3^2S_{\frac{1}{2}} (D_1)$ and $3^2P_{\frac{3}{2}} \rightarrow 3^2S_{\frac{1}{2}} (D_2)$. Calculate the Lande's g- factors for these levels. *(calculated in @)* (2012)
29. Discuss the fine structure of sodium D line. Draw D_1 and D_2 lines due to the transitions between 2P and 2S levels. *3p - (D₁) - 3s, 3p - (D₂) - 3s, $\lambda_{D_1} = 5896 \text{ \AA}$, $\lambda_{D_2} = 5890 \text{ \AA}$* (2013)
30. A sample of certain element is placed in a magnetic field of flux density 0.3 tesla. How far apart is the Zeeman component of a spectral line of wavelength 4500 A? Given: $e/m = 1.76 \times 10^{11} \text{ c/kg}$, $c = 3.0 \times 10^8 \text{ ms}^{-1}$. *Explain Zeeman effect $\rightarrow \Delta E = \mu_B B$ (for normal s=0) $\rightarrow \Delta \lambda = \frac{\lambda^2 e B}{4\pi mc}$, $\Delta E = hc \Delta \lambda$, $\Delta \lambda = \frac{\lambda^2 e B}{4\pi mc}$, $\Delta \lambda = 0.0255 \text{ \AA}$* (2014)
31. Obtain an expression for the normal Zeeman shift. Illustrate the Zeeman splitting of spectral lines of H atom and the allowed transitions for the $l = 1$ and $l = 2$ states. *Explain Zeeman effect $\rightarrow \Delta E = \mu_B B$ (for normal s=0) $\rightarrow \Delta \lambda = \frac{\lambda^2 e B}{4\pi mc}$, $\Delta E = hc \Delta \lambda$, $\Delta \lambda = \frac{\lambda^2 e B}{4\pi mc}$, $\Delta \lambda = 0.0255 \text{ \AA}$* (2014)
32. What is Zeeman Effect? How can it be understood on the basis of quantum mechanics? *(done)* (2015)
33. Obtain Zeeman splitting for sodium D-lines. *Define $\rightarrow$ types + selection rule, Conclude with (2015) linear Zeeman effect* (2015)
34. Find the magnetic moment of an atom in $3P_2$ state, assuming that LS coupling holds for this case. *$\mu_J = -\frac{e}{2m} g_J \vec{J}$ (assuming LS coupling) $\rightarrow$ solve all $\rightarrow$ conclude strong B, breaks coupling* (2015)
35. In the Stern-Gerlach experiment using Ag atoms, the oven temperature is 1000 K, $l \approx 25 \text{ cm}$ and $\frac{\partial B_z}{\partial z} \approx 10^3 \text{ Tesla/m}$. Calculate the separation of the two components. *Draw figure, Locate formula, $Z = \pm \mu_B (g_J) \frac{l^2}{2a k T} = 5.3 \times 10^{-3} \text{ m}$* (2016)
36. Describe Stern-Gerlach experiment. Discuss how it has explained space quantization and electron spin. Find the value of angle between the spin angular momentum S and its z-component of an electron moving along the external magnetic field B . *$\cos(\theta_S) = \frac{m_s}{\sqrt{s(s+1)}}$, $\theta_S = 18.4^\circ$* (2016)
37. The series limit wavelength of Balmer series in hydrogen spectrum is experimentally found to be 3646 A. Find the wavelength of the first line of this series. *[Series limit $\rightarrow$ shortest wavelength] $\frac{1}{\lambda_{\infty}} = R \left(\frac{1}{4} - \frac{1}{\infty} \right) \Rightarrow \lambda_{\infty} = 4/R$, $R = 1.097 \times 10^7 \text{ m}^{-1}$* (2016)
38. Compute the allowed spectral terms for two non-equivalent p-electrons of the basis of Pauli's exclusion principle. *for s=1, m_s values are same, same m_s values be avoided, as per Breit's scheme, Conclusion: 10, 15, 20, 25, 30, 35, 40, 45, 50, 55, 60, 65, 70, 75, 80, 85, 90, 95, 100, 105, 110, 115, 120, 125, 130, 135, 140, 145, 150, 155, 160, 165, 170, 175, 180, 185, 190, 195, 200, 205, 210, 215, 220, 225, 230, 235, 240, 245, 250, 255, 260, 265, 270, 275, 280, 285, 290, 295, 300, 305, 310, 315, 320, 325, 330, 335, 340, 345, 350, 355, 360, 365, 370, 375, 380, 385, 390, 395, 400, 405, 410, 415, 420, 425, 430, 435, 440, 445, 450, 455, 460, 465, 470, 475, 480, 485, 490, 495, 500, 505, 510, 515, 520, 525, 530, 535, 540, 545, 550, 555, 560, 565, 570, 575, 580, 585, 590, 595, 600, 605, 610, 615, 620, 625, 630, 635, 640, 645, 650, 655, 660, 665, 670, 675, 680, 685, 690, 695, 700, 705, 710, 715, 720, 725, 730, 735, 740, 745, 750, 755, 760, 765, 770, 775, 780, 785, 790, 795, 800, 805, 810, 815, 820, 825, 830, 835, 840, 845, 850, 855, 860, 865, 870, 875, 880, 885, 890, 895, 900, 905, 910, 915, 920, 925, 930, 935, 940, 945, 950, 955, 960, 965, 970, 975, 980, 985, 990, 995, 1000* (2016)
39. Explain in detail L-S coupling and j-j coupling schemes. *(already done)* (2016)
40. What is Lamb shift? What is its significance in determining the fine structure of H_α Balmer line in hydrogen atom? *(explained below)* (2016)
41. Calculate the radius of electron orbit for Li^{++} in ground state. *Hydrogen like, Bohr's model (quantization of L), write formula $r_n = a_0 \frac{n^2}{Z}$ ($Z=3$)* (2018)
42. Describe the importance of L-S and J-J coupling in atomic spectroscopy. What are experimental evidences of their existence? (2018)
43. What is Zeeman effect? Discuss the factors on which Larmor frequency is dependent. *Larmor, type, precession of J under influence of B about B direction, Larmor precession, $\omega_L = \frac{g \mu_B B}{\hbar}$, $\omega_L = \frac{g \mu_B B}{\hbar}$, $\omega_L = \frac{g \mu_B B}{\hbar}$, $\omega_L = \frac{g \mu_B B}{\hbar}$* (2018)

44. Discuss the fine structure of hydrogen spectrum. How is it of importance in the astronomical observations? (2018)

45. Define mathematically the Bohr radius of a hydrogen atom and who that the binding energy at state n of this atom can be given by

defn Bohr Model $\rightarrow$ nearly circular model + $L = n\hbar/2\pi \rightarrow e^-$ motion in stable orbit

putting v in eqn of r_n

$$r_n = \frac{4\pi\epsilon_0\hbar^2}{m_e e^2} \frac{n^2}{Z}$$

$\Rightarrow r_1 = a_0 = 0.529 \text{ \AA}$

$$E_n = -\frac{1}{2} \frac{Z^2 e^2}{(a/Z) 4\pi\epsilon_0 n^2} = -\frac{13.6}{n^2} \text{ eV}$$

$E_1 = -13.6 \text{ eV}$

$E_n = \frac{1}{2} m_e v^2 - \frac{Z e^2}{4\pi\epsilon_0 r_n}$

$\Rightarrow E_n = -\frac{Z e^2}{8\pi\epsilon_0 r_n}$

Conclude fine structure spin effect, but explained ground state

where Z is the atomic number of H atom. Calculate the numerical values of a and E_1 of H atom. (2019)

46. Describe normal and anomalous Zeeman effect. Explain how it lifts the degeneracy in hydrogen atom.

Zeeman effect is splitting of spectral line into several components ($\Delta m = 0, \pm 1$) under weak B

$\Delta m = 0 \rightarrow$ normal Zeeman effect

$\Delta m = \pm 1 \rightarrow$ anomalous Zeeman effect

$\Delta m = 0, \pm 1$ components

$\Delta m = 0 \rightarrow$ longitudinal

$\Delta m = \pm 1 \rightarrow$ transverse

$\Delta E = \mu_B B$

$g_j = \left[1 + \frac{j(j+1) + s(s-1) - l(l+1)}{2j(j+1)} \right]$

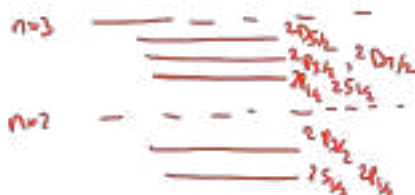
$\rightarrow$ Conclude with coupling breaks under strong B (Paschen Back)

47. What is lamb shift? Discuss its significance in determining the fine structure of H_α Balmer line in hydrogen atom. (2019)

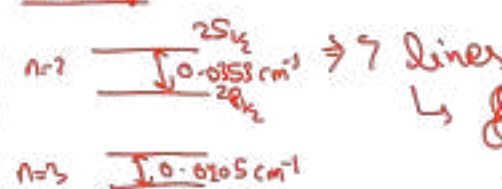
$$\Delta S = 0, \Delta J = 0, \pm 1, \Delta L = \pm 1$$

$\rightarrow$ 5 lines of H_α

as per Dirac's eqn



Lamb shift



7 lines



shaky e^- motion due to interaction of e^- with emp

for Deuterium at 1658 MHz & He at 14 GHz

$$\Delta E = hc \Delta \bar{\omega} \rightarrow \text{Zeeman shift}$$

$$\bar{\omega} \lambda = 1 \Rightarrow [\lambda \Delta \bar{\omega} = \lambda^2 \Delta \bar{\omega} \lambda]$$

$$\Delta \bar{\omega} = \frac{(e/m) \hbar B}{4\pi c}$$

for normal Zeeman effect

$\rightarrow$ in terms of Lorentz unit

Example: p^3 configuration [Main] [Q-22 Tut-4 (2008)]

$m_l = 1$	0	-1	1	0	-1	} of these, three electrons are taken for combination, $6C_3 = 20$ such combination	
$m_s = 1/2$	$1/2$	$1/2$	$-1/2$	$-1/2$	$-1/2$		
(a)	(b)	(c)	(d)	(e)	(f)		

(abc)	(abd)	(adb)	(abf)	(acb)	(ace)	(acf)	(ade)	(adf)	(aef)
$m_l = 0$	2	1	0	1	0	-1	2	1	0
$m_s = 3/2$	$1/2$	$1/2$	$1/2$	$1/2$	$1/2$	$1/2$	$1/2$	$1/2$	$1/2$
(bcd)	(bce)	(bce)	(bde)	(bde)	(bde)	(bde)	(cde)	(cde)	(cde)
$m_l = 0$	-1	-2	1	0	-1	0	-1	-2	0
$m_s = 3/2$	$1/2$	$1/2$	$1/2$	$1/2$	$1/2$	$1/2$	$1/2$	$1/2$	$1/2$

Highest values of m_l as 2 indicate a D term with $m_s = 1/2, -1/2 \Rightarrow S = 1/2$

Of remaining highest m_l value is 1, with $m_s = 1/2, -1/2 \Rightarrow 2P(L=1, S=1/2)$ terms

(remaining terms in box corresponds to $S = 3/2$) $\Rightarrow 4S$ term

$2P^o, 2D^o, 4S^o \rightarrow 4S$ is ground state

Example: $4p^2 5s$

two equivalent electrons & non-equivalent electrons.

$$4p^2 5s \Rightarrow 4p^2 + 5s \Rightarrow \left\{ \begin{array}{l} {}^1S + S = {}^2S \\ (L=0, S=0) (L=0, S=1/2) (L=0, S=1/2) \\ {}^3D + S = {}^4D \\ (L=2, S=1) (L=2, S=1/2) (L=2, S=1/2) \end{array} \right.$$

PHYSICS PAPER- II: MOLECULAR PHYSICS

TUTORIAL SHEET: 5

1. i. Discuss qualitatively the occurrence of rotational energy levels of diatomic molecule. Write down the selection rule
- ii. Is it possible to obtain pure rotational spectra of H₂, HF, O₂ and NO molecules? Can one obtain pure rotational spectra in emission? Comment. (On both of the above mentioned queries).
- iii. In CO molecule $J=0 \rightarrow J=1$ line occurs at frequency 1.153×10^{11} Hz. Calculate the moment of inertia of CO molecule.
- i. What is Raman Effect? How does it differ from Rayleigh scattering? Write down its important applications.
- ii. Explain Raman Effect with the aid of quantum mechanical theory.
- iii. Why is Raman Effect considered complimentary to infrared absorption spectra?
- iv. A Raman Stoke's line is observed at 552 nm. When a sample is excited by Hg green line of wave length 546 nm. Calculate the wave length of anti-stoke's line (2001)

2. i. A sample is irradiated by a 5000 Å radiation to give a Raman line at 5050.5 Å. Calculate the Raman frequency.
- ii. Explain, both red and violet degraded bands have been observed in electronic band systems, but rotation-vibration spectra show bands degraded to red only. (2002)

3. Calculate the moment of inertia of HCl molecule from the expression $= 20.68(J+1)$ cm where $J = 0, 1, 2, \dots$ (2002)

4. A diatomic gas is found to have a number of absorption bands in the ultraviolet region. Each band is observed to have a fine structure. How can the observed spectrum be explained and analysed? (2002)

5. Explain the nature, origin and significance of 21 cm Hydrogen line. (2002)

6. Calculate the change in rotational constant B_0 when deuterium is substituted for hydrogen in a hydrogen molecule. (2002)

7. (i) What are the typical energies (eV) of the radiations, required to excite (i) electronic transitions (ii) vibrational transitions and (iii) rotational transitions in a molecule?
- (ii) The rotational spectrum of HI consist of equidistant lines with a separation of 12.8 cm^{-1} . Calculate the (a) M.I. and (b) bond length of HI molecule. (2003)

8. Discuss the salient features of O, P, Q, R and S branches of electronic spectrum of diatomic molecules. Which spectrum contains only branches, both branches and bands, branches and progressions? (2003)

9. Show that a spinning nucleus processes in a magnetic field. Explain the underlying principle of NMR spectroscopy. The magnetic moment of a proton is $2.793 \mu_N$ Calculate the radio frequency at which nuclear magnetic resonance occurs in water kept in a magnetic field of 1.5 T. (2003)

10. Water vapor shows Infrared absorption bands at 1595 cm^{-1} , at 3652 cm^{-1} and at 3756 cm^{-1} . If the Raman spectrum is observed with the Argon ion laser 514.5 nm , at what wavelength would you expect to find the Raman - stokes lines? (2004)
11. High resolution laser spectroscopy can resolve lines that cannot be resolved by conventional spectroscopy. Justify this statement. (2004)
12. Briefly describe rotational fine structure of electronic spectra of diatomic molecules. (2004)
13. What are the selection rules for rotational spectra? Draw a neat diagram to show the allowed rotational transitions. (2004)
14. Explain the molecular phenomenon of spontaneous emission between two electronic states of the same multiplicity and differentiate it from that of different multiplicity. (2005)
15. The constants B and V_0 for a KCl molecule have values $143 \times 10^{-5}\text{ eV}$ and $84 \times 10^{12}\text{ s}^{-1}$ respectively. Determine the number of rotational levels between the vibrational levels $v=0$ and $v=1$. (2005)
16. How can the pure rotation spectrum of H_2 molecule be observed? If the bond length of H_2 molecule is 0.07417 nm . What would be the spacing of lines in its spectrum? (2005)
17. Differentiate between Rayleigh and Raman scatterings. Why is Raman scattering considered to be a breakthrough in molecular spectroscopy? What are the advantages of using laser light in Raman spectroscopy? (2005)
18. Obtain an expression for the vibration-rotation energy levels of a diatomic molecule in a given electronic state. The wave numbers of the vibrational transitions occurring in HF, HCl and HI molecules are 34143 cm^{-1} , 29889 cm^{-1} and 23095 cm^{-1} respectively. Compare the force constants of these three molecules. (2006)
19. (i) State and explain Franck – Condon Principle. (done)
(ii) In three separate pairs of electronic states (one ground and another upper state) of diatomic molecules, the average inter nuclear distances are-
(i) equal; (done)
(ii) slightly greater;
(iii) appreciably greater.
In accordance with Franck- Condon principle, obtain most favourable transitions from the ground state vibrational levels in each of the above situations.
Illustrate the transitions by drawing suitable energy – level diagrams. (2007)
20. Draw the potential energy of a diatomic molecule as a function of inter atomic distance. Mark the vibrational and rotational energy levels. Explain the selection rule for transition between vibrational states. (2008)

21. Assume a diatomic molecule consisting of two atoms of masses m_1 and m_2 , separated by a distance $\vec{r}$. Write down the Hamiltonian operator for the molecule. Determine the rotational energy levels of the molecule, (2008)
22. Write full form of acronyms EPR and NMR. Give underlying principle of EPR. (2008)
23. Explain fluorescence and phosphorescence in electronically excited molecules. (2009)
24. Discuss the vibrational spectra of a diatomic molecule treating it as a harmonic oscillator as well as an anharmonic oscillator and compare them. (2009)
25. Explain Born – Oppenheimer approximation. Discuss the intensity distribution in the vibrational electronic spectra of a diatomic molecule on the basis of Franck – Condon principle. (2009)
26. Calculate, giving necessary steps, the radio frequency at which nuclear magnetic resonance occurs in water kept in a uniform magnetic field of $2.4T$. The magnetic moment of proton is $2.793\mu_N$. (2010)
27. (i) Discuss occurrence of rotational energy levels of a diatomic molecule and show that the pure rotation spectrum of such a molecule consists of a series of equally spaced lines separated by a constant wave number difference $2B$. Write down the selection rules.
(ii) Is it possible to obtain pure rotational spectra of H_2 , HF, O_2 and NO molecules?
(iii) In CO molecule $J = 0 \rightarrow J = 1$ line occurs at a frequency 1.153×10^{11} Hz. Calculate the moment of inertia of CO molecule. (2010)
28. What is Raman Effect? How does it differ from Rayleigh scattering? Explain Raman Effect on the basis of quantum mechanical theory. How is Raman Effect experimentally studied? What are the advantages of using laser sources in the study of Raman Effect? (2010)
29. State and explain Franck-Condon principle. Discuss its applications in molecular spectroscopy. (2010)
30. On the basis of inertia I_A , I_B and I_C each about X, Y and Z axes respectively, how can you classify molecules? (2011)
31. (a) (i) Treating a diatomic molecule as a simple oscillator, obtain its energy (vibrational) levels.
(ii) The observed vibrational frequency of the CO molecule is 6.42×10^{13} Hz. What is the effective force constant of the molecule? ($C = 12u$, $O = 16u$)
(b) (i) Discuss pure rotational spectra of linear molecules
(ii) What is Lamb shift? (2011)
32. Why should rotational Raman spectrum show a separation of the first Raman line from the exciting line as $6B \text{ cm}^{-1}$. While the separation between successive lines is equal to $4B \text{ cm}^{-1}$, when B is the rotational constant? (2012)

33. What will happen to the energy level of an unpaired electron when subjected to an external magnetic field? Find the condition for electron spin resonance. (2013)
34. (a) With proper selection rules, construct the energy level diagram and allowed transitions for ESR spectrum of hydrogen atom. (b) Why are Raman active vibrations and infrared vibrations in CO_2 molecule complementary to each other? (c) In a Raman spectrum of linear triatomic molecule, the first three lines are 4.86 , 8.14 and 11.36cm^{-1} . Calculate the rotational constant, B and the moment of inertia of the molecule. (Given $h = 6.626 \times 10^{-27}\text{J.s}$, $C = 3.0 \times 10^{10}\text{cm/sec.}$) (2013)
35. Discuss the vibrational spectra of a diatomic molecule treating it as an anharmonic oscillator. (2014)
36. Explain how the nuclear spin I depends on the mass number A and atomic number Z of atoms. (2014)
37. (i) Obtain an expression for the resonance condition in NMR.
(ii) Explain the relaxation processes in NMR spectroscopy. (2014)
38. Establish that: $hc = 1240\text{ MeV. fm}$
*Info $\rightarrow h$ is Planck constant
 c is speed of light*

$$hc = \frac{6.626 \times 10^{-34} \times 3 \times 10^8}{1.6} \text{ eV-m} = 1242 \text{ eV-nm}$$
(for 1 nm wavelength, 1240 eV is associated) (2015)
39. Hydrogen molecule is diatomic. Obtain the rotational energy levels of this molecule. Write down the selection rules. Obtain the smallest energy required to excite the lowest rotational mode. (2015)
40. The observed vibrational frequency of CO molecule is $6.42 \times 10^{13}\text{ Hz}$. What is the effective force constant of the molecule?
Assuming SHO motion associated with molecule,

$$\omega_{\text{eff}} = \frac{1}{2\pi} \sqrt{\frac{K}{\mu}}$$

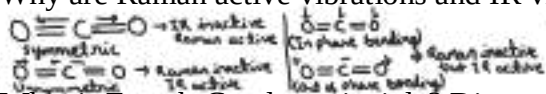
$$K = 4\pi^2 \nu^2 \times \mu_{\text{CO}} \quad \& \quad \mu_{\text{CO}} = \frac{12 \times 16}{12+16} \text{ amu}$$
 (2015)
41. Show that the lines in the absorption spectra corresponding to the rotational transitions from two adjacent energy levels of a medium sized molecule at room temperature have comparable intensities. (2017)
42. Given the force constant of HCl molecule = 516 Nm^{-1} , determine the wave number of the fundamental mode of vibration of the molecule. How many transition lines one can expect in the vibration spectra of HCl molecule at room temperature? (2017)
43. Explain Stokes and anti-Stokes Raman scattering with the help of energy level diagram. For a diatomic molecule, obtain expressions for transition energies of its Raman spectra with rotational fine structure and hence the wave numbers of the Stokes lines. (2017)
44. Explain why lines in some Raman spectra are found to be plane polarized to different extents even though the exciting radiation is completely unpolarized? (2017)
45. State Franck-Condon principle. Define Franck-Condon factors. Using schematic diagram, explain the decay of excited states leading to the phenomena of fluorescence and phosphorescence. (2017)
46. Explain the principle of Nuclear Magnetic Resonance (NMR) with the help of an energy level diagram. Give examples of nuclei which exhibit NMR. What major inferences can be drawn from an NMR spectra? (2017)

47. In an NMR experiment, hydrogen atoms are subjected to a magnetic field of 5.0 T. Determine the difference in energy (kJ/mol) between two spin states of the nuclei of hydrogen atom and the frequency of radiation required for NMR. (2017)

48. What is nuclear precession? How is it used in the principle of working of NMR? (2018)

49. Discuss the theory of rotational and vibrational spectra of diatomic molecules. What is the difference between fluorescence and phosphorescence? (2018)

50. Why are Raman active vibrations and IR vibrations in CO_2 molecule complementary to each other? (2019)



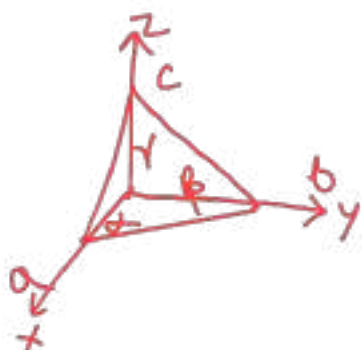
change in permanent dipole moment

CO_2 has inversion centre with centre of symmetry $\rightarrow$ laws of mutual exclusion $\rightarrow$ complementarity of IR and Raman of the vibrational modes $\rightarrow$ symmetry

51. What is Franck-Condon principle? Discuss the intensity distribution in the vibrational electronic spectra of a diatomic molecule on the basis of this principle. (already done) (2019)

Born-Oppenheimer Approximation is the assumption that the electronic motion and the nuclear motion in molecules can be separated. It leads to a molecular wave function in terms of electron positions and nuclear positions.

Assumptions: 1. The electronic wavefunction depends upon the nuclear positions but not upon their velocities, i.e., the nuclear motion is so much slower than electron motion that they can be considered to be fixed. 2. The nuclear motion (e.g., rotation, vibration) sees a smeared out potential from the speedy electrons



$$h = \frac{a}{A}, k = \frac{b}{B}$$

$$[h:k:l] = \left[\frac{a}{A}, \frac{b}{B}, \frac{c}{C} \right]$$

$$\alpha = \cos^{-1} \left(\frac{a}{r} \right) = \frac{a}{h} \quad \beta = \frac{b}{r} = \frac{b}{k}$$

TUTORIAL SHEET: 6 NUCLEAR PHYSICS

[Please check Notebook too]

1. Explain why a neutral particle such as neutron possesses a finite value of magnetic moment, and how it is determined? (2002)

2. Write down the configuration of ${}^7\text{Li}$, ${}^{13}\text{C}$ and ${}^{25}\text{Mg}$ in the ground state of the nuclear shell model. (2002)

3. Explain how neutrino was discovered from the β -decay of radioactive nuclei. (2002)

4. Explain shell model of the nucleus. Give evidence for nuclear shell structure. What are the limitations of the shell model? (2002)

5. Sketch carefully the binding energy per nucleon curve for stable nuclei. Explain its salient features. On the basis of this curve explain why fusion is possible only for low mass nuclei, where takes place in heavy nuclei. (2003)

6. List some of the important properties of deuteron and show that it is a loosely bound system only one bound state. (2003)

7. Distinguish between a nuclear reaction and decay. Which conservation laws are obeyed in nuclear reactions? Explain the significance of Q value. Calculate the Q value for the following nuclear reaction, ${}^2\text{H} + {}^1\text{H} \rightarrow {}^3\text{He} + n$ (2003)

8. Calculate the fission rate of ${}^{235}\text{U}$ required to produce 4 watts and the amount fissioning of 1 kg of ${}^{235}\text{U}$. (Assuming energy released per fission of ${}^{235}\text{U}$ is 200 Mev). (2004)

9. Calculate Binding energy per nucleon of ${}^4_2\text{He}$ and ${}^{16}_8\text{O}$. (Draws curve of $B.E.$ vs A) (2004)

10. Write down the Bethe-Weizsacker formula. Explain various terms of the formula. (2004)

11. Explain how energy is released in stars. (2004)

12. Describe in detail one experiment for detection of Neutrino. (2004)

13. What is the difference between neutrino and antineutrino? How is helicity of neutrino determined? (2004)

14. (a) Write the Weizsacker mass formula and explain the significance of various terms.

- (b) Determine the amount of ${}^{210}_{84}\text{Po}$ necessary to provide a source of ${}^{210}_{84}\text{Po}$ α -particles of strength 5 mCi. The half-life of is 138 days. (2005)

15. Describe how parity violation was experimentally detected in ${}^{60}\text{Co}$ β -decay. How was the observed asymmetry in the distribution of emitted electrons explained? (2005)

Single Particle shell Model (asked in Main) [TSAG @]

Assuming QM treatment of nuclei, and assuming unpaired nucleon contribute alone to spin, parity & moment, it assumes nucleon move nearly freely under influence of averaged effect of rest $(A-1)$ nucleons.

We also assume nucleon imagined to move in a shell depending on energy & angular momenta. (quantized) analogous to atomic electrons.

Now we use spin-orbit coupling b/w nucleons [s & s combine to give j's & j's couple to give Jtot].

$$W = -\lambda \frac{1}{Mc^2} \left(\frac{\partial U}{\partial r} \right) \vec{L} \cdot \vec{S} \quad \left[\vec{L} \cdot \vec{S} = \frac{L^2 + S^2 - J^2}{2} \right]$$

Now putting $U = V_0 + \frac{1}{2} m \omega^2 r^2$;

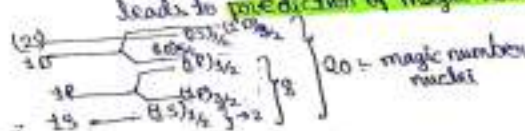
$$E_{n\ell} = \hbar \omega \left(N + \frac{3}{2} \right) \quad (\text{where } N = 2n + \ell - 2)$$

for each value of N , each shell is occupied by $N_A = (N+1)(N+2)$

N	$N_{max} = \sum N_A = \sum (N+1)(N+2)$
0	2
1	8
2	20
3	40
4	70
5	112

not stable nuclei

To correct this, one incorporates spin-orbit interaction b/w nucleons and centrifugal term = $\frac{\hbar^2 \ell(\ell+1)}{2mr^2}$ which leads to prediction of magic numbers.



Significance of Gellmann-Nishijima Relation

$$Q = I_z + \frac{1}{2}(B+S)$$

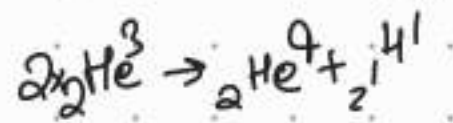
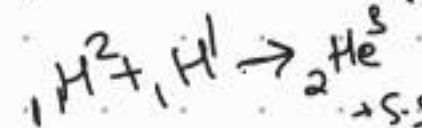
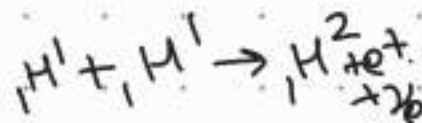
- useful in explaining change of Kaon ~~into~~ mesons by introducing concept of strangeness → conserved in production & violated in decay.
- explains subparticles with same B & isospin ($2I+1$ such sub particles). [postulates analogous to em field, charge gives rise $(B+1)$ particles like Zee-man effects]

$$\therefore Q = a I_z + b \Rightarrow \langle Q \rangle = b \quad (\text{as } \langle I_z \rangle = 0 \text{ as } I_z \text{ goes from } -I \text{ to } +I)$$

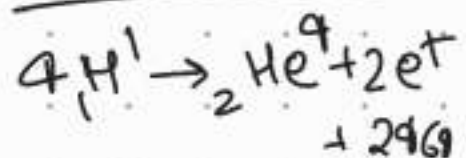
$$(a = e \text{ \& } b = \langle Q \rangle = e \frac{Y}{2}) \Rightarrow Q = e \left(I_z + \frac{Y}{2} \right) \rightarrow \text{modified relation,}$$

(Since B is always conserved, conservation of Y tells us about strangeness)

Neutron stars are formed when a massive star runs out of fuel and collapses. The very central region of the star - the core - collapses, making every proton & electron into a neutron. (collapse occurs as after iron, sudden stoppage of energy generation causes the core to collapse and the outer layers of the star to fall onto the core.) Gravity constantly works to try and cause the star to collapse. The star's core, however is very hot which creates pressure within the gas. This pressure counteracts the force of gravity, putting the star into what is called hydrostatic equilibrium. When the Sun runs out of hydrogen to fuse, the balance tips in the favor of gravity, and the star starts to collapse. But compacting a star causes it to heat up again and it is able fuse what little hydrogen remains in a shell wrapped around its core. This burning shell of hydrogen expands the outer layers of the star. When this happens, our Sun will become a red giant; it will be so big that Mercury will be completely swallowed. Most neutron stars are observed as pulsars. Pulsars are rotating neutron stars observed to have pulses of radiation at very regular intervals that typically range from milliseconds to seconds. A pulsar (from pulse and -ar as in quasar) [1] is a highly magnetized rotating compact star (usually neutron stars but also white dwarfs) that emits beams of electromagnetic radiation out of its magnetic poles.



+ 12.8 MeV



- (a) Differentiate between K-capture and inverse β -decay. *used in detectⁿ of neutrino by concn. neutrinos*
(b) Explain what is internal conversion and how it differs from β -decay. (2005)

16. Calculate the spin and parities of the ground states of ${}^4_2\text{He}$ and ${}^{67}_{30}\text{Zn}$ nuclei. (2005)

defⁿ shell model $\rightarrow$ write spin & parities $\rightarrow$ success & limitations $\rightarrow$ diagram of shell structure

17. Calculate the Q - value of the reaction : ${}^9_4\text{Be}({}^4_2\text{He}, n){}^{12}_6\text{C}$

Given

$$\text{Mass}({}^9_4\text{Be}) = 9.012183u$$

$$\text{Mass}({}^4_2\text{He}) = 4.002603u$$

$$\text{Mass}({}^{12}_6\text{C}) = 12.0000u$$

$$Q = [(m_{\text{Be}} + m_{\text{He}}) - (m_{\text{C}} + m_{\text{n}})]c^2$$

(2007)

18. Write down the nucleonic configuration of ${}^7_3\text{Li}$, ${}^{12}_6\text{C}$, ${}^{17}_8\text{O}$ and ${}^{27}_{13}\text{Al}$ in the ground state of the nuclear shell model and hence calculate the corresponding ground state angular momenta and parities. How do the observed ground state angular momenta and parities agree with those predicted on the basis of the shell model? *[Repeated in Q # 38]* (2007)

19. (i) Write down the Bethe - Weizsacker semi-empirical mass formula for a nucleus. Explain the significance of each term occurring in it. Discuss the stability of a nucleus against β -decay. What is the effect of pairing term on stability? *(M = aA - bZ^2/A^{1/3} - cZ(Z-1)/A^2 - dA^{-1/2} + \delta)* *(Z = 40 for 91)*

- (ii) What are mirror nuclei? How does the charge independence of nuclear force emerge from this concept? *(done already)* (2007)

20. What is β -decay? Give three examples of β -decay of nuclei. Explain how neutrino hypothesis helped in explaining conservation of energy- momentum and angular momentum. What is Kurie plot? How can one use Kurie plot to set limits on mass of the neutrino? *(done already)* (2008)

21. What is nuclear isomerism? Give two examples and explain how this can be understood from single particle shell model. *existence of isobaric, isotopic nuclei in diff. energy levels* (2008)

22. Write down the Yukawa potential and derive an expression for the mass of π -meson in terms of range of nuclear force. What part of the Nucleon - Nucleon interaction is explained by this potential? *strong force nuclear force* (2009)

23. What is the simplest two-nucleon bound system? How does its study help in obtaining information about nuclear forces? Indicate clearly how the non-central forces can be explained by the observed magnetic moment of this system. *Heavier $\neq$ (1/2 spin) neutron + (1/2 spin) proton $\rightarrow$ suggesting mixed state $\rightarrow$ $\Delta \neq 0 \rightarrow$ non-central force* (2009)

24. Compare the methods of electron scattering and neutron scattering experiments to obtain information about nuclear size. *electromagnetic charge distribution for size of nuclei* *nuclear force based cross section to calculate size* (2009)

25. Calculate the packing fraction and the binding energy per nucleon for ${}^{16}_8\text{O}$ and ${}^{87}_{38}\text{Sr}$ nuclei. (2010)

defⁿ B.E.

$$P.F. = \left(\frac{M}{A} - 1 \right) \text{ where } M = \text{mass of nuclei}$$

($\downarrow$ P.F. $\Rightarrow$ $\uparrow$ stability)

$$B.E. = [m_p Z + m_n (A - Z)] - M$$

$\rightarrow$ B.E. gives stability of nuclei & mass-energy equivalence

P-P cycle & C-N cycle enable for fusion → star's energy source

26. A star converts all its hydrogen to helium, achieving 100% helium. It then converts the helium to carbon via the reaction $4_2\text{He} + 4_2\text{He} + 4_2\text{He} \rightarrow {}^{12}_6\text{C} + 7.27\text{MeV}$. The mass of the star is $5.0 \times 10^{32}\text{ kg}$ and it generates energy at the rate of $5 \times 10^{30}\text{ kg}$. How long will it take to convert all Helium to carbon at this rate? (2010)

$12\text{ amu} \rightarrow 7.27\text{ MeV}$
 $\dot{t} = \frac{E}{P} = \frac{5 \times 10^{30} \times 7.27}{12 \times 1.6 \times 10^{-13} \times 10^3} = \frac{3.635 \times 10^{43}}{1.92 \times 10^{16}} = 1.89 \times 10^{27}\text{ second} \rightarrow \text{takes years}$
 $P = \frac{2.9 \times 10^{13}\text{ MeV/second}}$

27. What are magic numbers? Discuss shell structure of a nucleus. How is this model able to explain various properties of nuclei? Discuss the limitations of this model. (2010)

→ defⁿ (already done) → conclude → significance & limitations

28. Predict from the single particle shell model the shell configuration, ground state spin and parity for the following nuclei: ${}^{13}_7\text{N}$, ${}^{17}_8\text{O}$. (2010)

$(1s_{1/2})^2 (1p_{3/2})^4 (1p_{1/2})^1 \rightarrow \text{spin} = 5/2; \text{parity} = -1$

29. Explain parity violation in β -decay. Describe how parity violation was experimentally detected in the decay of ${}^{60}\text{Co}$. (2010)

(done already)

30. Find the total kinetic energy of electron and antielectron neutrino emitted in beta decay of free neutron. (The neutron-proton mass difference is 1.30 MeV and mass of electron is 0.51 MeV) (2010)

$n \rightarrow p + e^- + \bar{\nu}_e$
 $(K.E.)_{e^- + \bar{\nu}_e} = (m_n - (m_p + m_e)) = 0.79\text{ MeV}$

→ conservation of energy

31. What are chain reactions? What do you mean by critical size of the core in which chain reaction takes place? What is critical mass? (2010)

K depends on conc. of nuclei, geometry of nuclei, material of nucleus

(min. mass for fusion when $K > 1$)

chain reactions → P-P cycle → $K = \text{neff}$
 Application: Nuclear reactor, while doing star

(K = multiplicative factor)

32. ${}^{235}\text{U}$ yields two fragments of $A = 95$ and $A = 140$. Obtain the energy distribution of the fission products. Assume that the two fragments are ejected with equal and opposite momentum. (2010)

(done already)

33. Show that nucleus is a quantum system. (2011)

→ shell model / space quantization / discrete energy level
 ↳ follows quantum mechanics / NMR
 ↳ simplest nuclear model explains nature of nuclear force

34. (a) (i) What is the importance of study of deuteron? Obtain the solution of Schrodinger equation for ground state of deuteron and show that deuteron is a loosely bound system.

(ii) What do you mean by non-central forces? (Tensor, $\Delta \neq 0$)
 $\frac{51d}{K} = \frac{1}{\pi^2 \hbar^2 m v_0} \Rightarrow 2.3 \Rightarrow R \approx 2.1\text{ fm}$

- (b) (i) What are chain reactions? What do you mean by critical size of the core in which chain reactions take place? What is critical mass? (series of self sustaining uncontrolled reaction)

- (ii) ${}^{235}\text{U}$ yields fragments of $A = 95$ and $A = 140$. Obtain the energy distribution of the fission products. Assume that the two fragments are ejected with equal and opposite momenta. (2011)

$[p_1 = p_2 \Rightarrow \frac{K_1}{m_1} = \frac{K_2}{m_2}]$
 $K.E. \propto \text{vel. yield}$

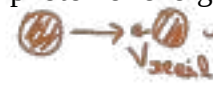
35. (a) Write down the Bethe-Weizsacker Semi-empirical mass formula for a nucleus. Explain the significance of each term. Discuss the stability of a nucleus against β -decay. What is the effect of pairing term on stability? (done already)

- (b) What are mirror nuclei? How does the charge independence of nuclear force emerges from this concept? ($\Delta B.E. = \Delta E_c$ (liquid drop) → $p-p \equiv n-n$ → charge symmetry) (2012)

↳ nuclear force is charge independent

36. Calculate the recoil energy of ${}^{26}_{11}\text{Fe}$ nucleus when it emits a gamma photon of energy 14 KeV. (2012)

$p_{\text{nuclei}} = p_\gamma = \frac{E_\gamma}{c} \Rightarrow E_{\text{recoil}} = \frac{E_\gamma^2}{2mc^2} = 4.89\text{ meV}$



37. Calculate the Q-value of the reaction: ${}^9_4\text{Be}({}^4_2\text{He}, n){}^{12}_6\text{C}$

Given:

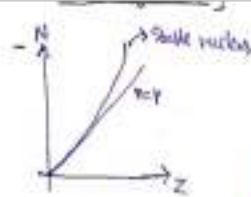
(done already)

$$\text{Mass } \left({}^9_4\text{Be} \right) = 9.01283 \text{ u}$$

$$\text{Mass } \left({}^4_2\text{He} \right) = 4.002603 \text{ u}$$

$$\text{Mass } \left({}^{12}_6\text{C} \right) = 12.000 \text{ u}$$

$$\text{Mass } (n) = 1.0086652 \text{ u}$$



When nucleus $\uparrow$, radius $\uparrow$
to keep it stable, one needs strong force
force and there is also $\uparrow$ in radiative
repulsion, so, Neutrons $\uparrow$ to offset it

013)

38. Write down the nucleonic configuration of, ${}^7\text{Li}$, ${}^{12}\text{C}$, ${}^{17}\text{O}$, and ${}^{27}\text{Al}$ in the ground state of the nuclear shell model and hence calculate the corresponding ground state angular momenta and parities. How do the observed ground state angular momenta and parities agree with those predicted on the basis of shell model? (2013)

39. What is the role of neutrino in the weak interaction of radioactive nuclides? Explain the experimental detection of neutrino. (2013)

40. Explain parity violation in β -decay. Describe how parity violation was experimentally detected in the decay of ${}^{60}\text{Co}$. Mention any other decay process in which the parity violation has been demonstrated. (2013)

41. Explain why stable light nuclei have equal number of protons and neutrons whereas heavy nuclei have excess of neutron. (2014)

42. What are the magic numbers in nuclei? List the experimental evidences indicating their existence. (2014)

43. It is possible to estimate the nuclear radius from the study of alpha decay? Explain how. (2014)

44. State the basic assumption of single particle shell model. How do the centrifugal and spin-orbit terms remove the degeneracy of a three-dimensional spherical harmonic oscillator? (2016)

45. Predict the spin and parity of ground states of the following nuclei on the basis of shell model:

(i) $8^{O^{15}}$

(ii) $8^{O^{16}}$

(iii) $17^{Cl^{35}}$

(2016)

46. Explain why the deuteron has no excited state. (2016)

47. Estimate the order of nuclear radius of lead ($Z = 82$) using the large angle (back) scattering of alpha particles of energy 10 MeV incident on a target (lead). (2017)

[Given : $(4\pi \epsilon_0)^{-1} = 9 \times 10^9 \text{ Nm}^2 \text{C}^{-2}$]

$Z = 82$ for lead

27



Assuming back scattering $\theta \sim 180^\circ$
$$\frac{2Ze^2}{4\pi\epsilon_0 r_{\min}} = (K.E.)_\alpha$$

Conclusion $\rightarrow$ this gives approximate picture, accurate value can be obtained using nuclear force.

48. Distinguish between charge independence and charge symmetry of nuclear force. Give one example of each of these. (solution in text box) (2017)

49. Describe briefly how parity violation in β - decay was experimentally observed? What do you understand by the statement, 'neutrinos are left-handed'? (2017)

Helicity is pseudoscalar which changes during parity violation.

momentum & spin are in opposite direction

50. Given that the deuteron magnetic moment operator (in units of nuclear magneton) can be expressed as

Deuteron is triplet state $\Rightarrow \vec{\sigma}_n = \vec{\sigma}_p$

$$\vec{\mu}_d = \mu_n \vec{\sigma}_n + \mu_p \vec{\sigma}_p + \frac{1}{2} \vec{l},$$

Where $\vec{l}$ is the relative angular momentum between neutron and proton $\vec{\sigma}_n$ and $\vec{\sigma}_p$ are the Pauli spin operators and μ_n and μ_p are the respective magnetic moments. Find out the D-state probability of deuteron wave function. (done already)

[Given : $\mu_d = 0.857 \mu_N$, $\mu_n = -1.913 \mu_N$ and $\mu_p = 2.793 \mu_N$; μ_N (nuclear magneton)] (2017)

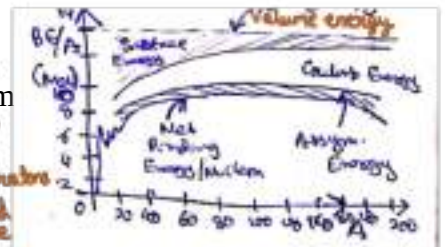
$$M(Z, A) = (m_p Z + (A - Z)m_n) - (B, E)$$

51. Write the semi-empirical mass formula pointing out the role of volume coulomb and symmetry energy correction terms.

$$B, E = a_1 A - a_2 A^{2/3} - a_3 A^{-1} Z(Z-1) - a_4 A^{-1} (A - 2Z)^2 - a_5 A^{-1/2}$$

($a_1 = 15.5$ MeV, $a_2 = 17$ MeV, $a_3 = 0.71$ MeV, $a_4 = 0.6$ MeV, $a_5 = 12$ MeV)

$a_1 \rightarrow$ excess nucleus
 $a_2 \rightarrow$ less neighbours less stable
 $a_3 \rightarrow$ repulsion between protons
 $a_4 \rightarrow$ odd-odd unstable nucleus, even stable



52. Draw a schematic diagram of the single particle energy levels in a shell model including the effect of spin-orbit coupling. Show how it explains magic numbers in nuclei. Give two examples to show how this scheme predicts the spins and parities of odd A nuclei. (2017)

$8O^{17} \Rightarrow$ spin = $\frac{5}{2}$, parity = +1 & $7N^{15} \Rightarrow$ spin = $\frac{1}{2}$ & parity = -1

53. Write the semi-empirical mass formula for nuclei and on its basis draw mass parabolas for odd and even isobars. What would be the most stable isobar in each case? (done already) (2018)

54. Obtain an expression for the magnetic moment of a nucleus having one nucleon outside the closed core. Use this to calculate the magnetic moment of $8O^{17}$ nucleus. (2018)

$$\mu_i = \frac{e\hbar}{4\pi m_p} \frac{j}{j+1/2} \Rightarrow \mu_i \text{ due to neutron } [(\frac{1}{2} \hbar)^2 (\frac{1}{2} \hbar)^2 (10 \hbar)^2] = [\frac{1}{2} \hbar, j = \frac{5}{2}]$$

55. Nuclear forces are mediated by exchange of π -mesons of rest mass 140 MeV. Estimate the range of nuclear forces. (2018)

$$R = \frac{c}{\Delta E} \Delta E \sim m_{\pi} c^2 \quad \& \quad \Delta E \sim \frac{\hbar}{R}$$

$$R \sim \frac{\hbar}{2 m_{\pi} c}$$

$$R \sim \frac{1}{4\pi} \times \frac{1240 \text{ eV} \cdot \text{nm}}{140 \text{ MeV}}$$

56. Assuming that the neutron-proton interaction has a square well form

$$V(r) = -V_0 \text{ for } r \leq b \\ = 0 \text{ for } r > b$$

deuteron nucleus is given as

$$\psi(r) = A \sin kr \text{ for } r \leq b$$

Magnetic moment of Deuteron

$\vec{\sigma}_n, \vec{\sigma}_p$ = Pauli spin operators

$$\vec{\mu}_d = \mu_n \vec{\sigma}_n + \mu_p \vec{\sigma}_p + \vec{L}_p \quad (\text{in terms of } \vec{\mu}_n) \quad \left\{ \vec{L}_p = \frac{\vec{L}}{2} \text{ in c.m. frame} \right\}$$

In COM frame

$$\vec{\mu}_d = \left(\frac{\mu_n + \mu_p}{2} \right) (\vec{\sigma}_n + \vec{\sigma}_p) + \left(\frac{\mu_n - \mu_p}{2} \right) (\vec{\sigma}_n - \vec{\sigma}_p) + \frac{\vec{L}}{2} \quad (\text{for triplet state } \vec{\sigma}_n = \vec{\sigma}_p)$$

$$\therefore \vec{\mu}_d = (\mu_n + \mu_p) \vec{S} + \frac{\vec{L}}{2} = (\mu_n + \mu_p) \vec{I} - (\mu_n + \mu_p) \frac{\vec{L}}{2} + \frac{\vec{L}}{2} \quad (\because \vec{I} = \vec{S} + \vec{S})$$

$\vec{I}$ & $\vec{L}$ is in z-direction

$$\langle \mu_d \rangle = (\mu_n + \mu_p) \langle I_z \rangle - (\mu_n + \mu_p) \frac{1}{2} \langle L_z \rangle \quad (i)$$

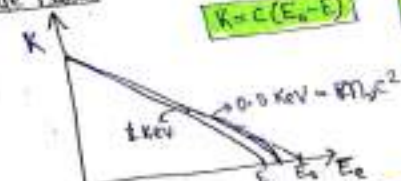
$$\left\{ \begin{aligned} \langle I_z \rangle &= \left\langle \frac{(\vec{I} \cdot \vec{I}) I_z}{I^2} \right\rangle \Rightarrow \langle L_z \rangle = \frac{L(L+1) + I(I+1) - S(S+1)}{2I(I+1)} \langle I_z \rangle \end{aligned} \right\}$$

$$\therefore I=1, S=1; \quad \langle L_z \rangle = \frac{L(L+1)}{4} I_z \quad (I_z=1) \quad \{ \langle L_z \rangle_S = 1/2 \times 0 = 0 \}$$

$$\{ |\psi\rangle = C_3 |1\rangle + C_0 |0\rangle, L=2 \text{ for D} \} \Rightarrow \langle L_z \rangle = \frac{6 |C_0|^2}{4} = \frac{3}{2} P_0 \quad (\text{as } P_0 = |C_0|^2)$$

Putting it in (i) gives, $\mu_d = (\mu_n + \mu_p) - (\mu_n + \mu_p) \frac{1}{2} \frac{3}{2} P_0 \Rightarrow P_0 \approx 0.04$

Klein Plot



$$K = C(E_0 - E)$$

$$K = \sqrt{\frac{P(E)}{F(E, E_0) P_0}}$$

where $P(E)$ is

- 1. $F(E, E_0)$ is form factor
- 2. P_0 is momenta of e-
- 3. $F(E, E_0) = 1$ for $Z=0$

$$Q_{\max} = E_e + E_{\bar{\nu}_e}$$

$$\text{to } E_0 [p \rightarrow n + e^- + \bar{\nu}_e]$$

$$\therefore E_0 = E_e + E_{\bar{\nu}_e} + m_e c^2 \quad \& \quad P_0 = \frac{1}{c} [E_0 (E_0 + 2m_e c^2)]^{1/2}$$

$$K = C(E_0 - E) [E_0 (E_0 + 2m_e c^2)]^{1/2} \quad \text{for } m_e = 0$$

m_e can be calculated from deviation (due to neutrino mass) from straight line near $E_0 = E_e$. $\Rightarrow$ deviation shift gives upper limit on m_e .

$\rightarrow$ Deuteron magnetic moment in 3S_1 state

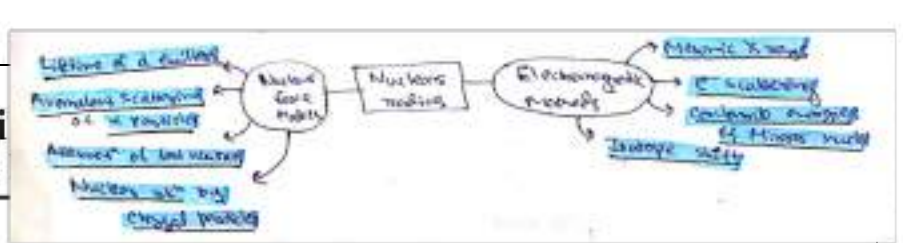
$$\left[\vec{\mu}_d = \left[(g_p \vec{L} + g_s \vec{S}) \cdot \vec{J} \right] \frac{\mu_N}{J(J+1)} \right] \quad \text{for } l=0; \quad \vec{\mu} = \vec{\mu}_p + \vec{\mu}_{\text{neutron}} \quad \mu_d = 0.8798 \mu_N$$

$$\vec{\mu}_{\text{spin}} = P_S \times \vec{\mu}_S + P_D \times \vec{\mu}_D$$

$$\vec{\mu}_{\text{spin}} = \left(\frac{g_p + g_n}{2} \right) \vec{S}$$

in com frame,

$$\frac{\vec{\mu}}{\mu_N} = \left[\frac{\vec{L} \cdot \vec{J}}{J(J+1)} + \frac{1}{2} (g_p + g_n) \frac{\vec{S} \cdot \vec{J}}{J(J+1)} \right] \vec{J}$$



$$= Ce^{-\gamma r} \text{ for } r > b$$

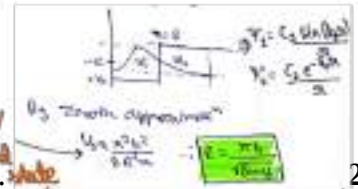
Where $k = \sqrt{\frac{M}{\hbar^2} (V_0 + W)}$ and $\gamma = \sqrt{\frac{MW}{\hbar^2}}$

Here M is the nucleon mass, W is the binding energy of deuteron and A and C are constants.

(i) Show that for a just bound state of deuteron

$$V_0 b^2 = \frac{\pi^2 \hbar^2}{4M}$$

[Schrödinger → boundary condition → Calculated R]
for excited state, $[x > 3\pi/2] \rightarrow V_0 R^2 \geq \frac{9\pi^2 \hbar^2}{4M} \rightarrow V_0 \geq 270 \text{ MeV}$
No excited state



(ii) Explain why deuteron is a loosely bound extended structure.

(2018)

57. Explain the various methods of finding the size of the nucleus. How will you determine the nuclear radius from the observation of beta rays resulting from nuclear transition when the initial and final nuclei are mirror nuclei?

$$\Delta E_c = \Delta BE = \frac{a_3(2Z \pm 1)}{A^{2/3}} = \frac{3e^2}{20\pi\epsilon_0 R_0 A^{2/3}} \quad (2019)$$

$$\Delta BE = (\Delta M_{\text{obs}} - (m_p - m_n))$$

58. Given that the single particle energy separation between $1d_{5/2}$ and $1d_{3/2}$ in ^{17}O is 5 MeV . Calculate the strength of spin-orbit interaction. It is observed that $1d_{5/2}$ level is lower than $1d_{3/2}$ level. (2019)

spin-orbit interaction defⁿ → $(\Delta E)_{LS} = -\alpha (\vec{L} \cdot \vec{S})$ $\{\vec{L} \cdot \vec{S} = (\frac{J^2 - L^2 - S^2}{2})\}$ $\alpha \rightarrow$ strength characteristics of atom.

59. Why is it not possible to detect the parity violation in weak interaction by observing only the beta decay rate? Justify your answer. (2019)

→ because to detect parity violation, one needs to show directⁿ in which β -decay is more or less change in helicity.
→ this explains existence of neutrinos.

draw diagram mentioned in Q → 40

(Q → 40)

nuclear force has "charge symmetry," which means that we can make an overall switch between protons and neutrons without changing forces among them. For instance, nn & pp scattering are the same (except for the obvious difference due to the electric charge). For example, "mirror nuclei," which are related by switching protons and neutrons, have very similar excitation spectra. Examples include ^{13}C and ^{13}N , ^{17}O and ^{17}F , etc.

A stronger version of the charge symmetry is "charge independence." Not only nn and pp scattering are the same, but also np scattering is also the same under the "same configuration", specified using the concept of isospin.

TUTORIAL SHEET: 7 PARTICLE PHYSICS

1. Illustrate with any five examples that production of strange particles conserves strangeness, parity and isotopic spin while their decay violates all these quantities. (2001)
2. Discuss the significance of Gellmann-Nishijima relation in strange particle production and decay. Give examples in support of your answer. Obtain quark quantum numbers using the modified relation for quarks. (Page 57 of Notes) (2001)
3. Which conservation laws are obeyed in nuclear reactions? Explain the significance of Q value of a nuclear reaction. (2002)
4. Distinguish between particle and antiparticle. How antiparticle were discovered? (2002)
Distac experiment
same mass, spin & isospin
5. Mention the conservation laws for strong, electromagnetic and weak interactions. (2002)
6. Give the characteristic properties of strange particles which distinguish them from non strange ones. Write the Gellmann – Nishijima relation and show how it is used for the classification of elementary particles. (2003)
7. State the conservation law which is violated in each of the following processes:
 (i) $n = p + \bar{e} + \nu$ (L_e , spin violated)
 (ii) $\pi^- + p = k^+ + k^-$ (B , spin not conserved)
 (iii) $\pi^- + p = \Sigma^+ + k^-$ (I_z , spin not conserved)
 (iv) $\Sigma^+ + k^- = p + p$ (B , spin, I_z non-conserved) (2003)
8. Which of the following reactions are allowed and forbidden under the conservation of strangeness, baryon number and charge conservation?
 (i) $\pi^+ + n = \lambda^0 + k^+$ (allowed)
 (ii) $\pi^+ + n = k^0 + k^+$ (allowed)
 (iii) $\pi^+ + p = \lambda^0 + k^+$ (allowed)
 (iv) $\pi^+ + p = \pi^0 + \lambda^+$ (allowed)
 (2004)
9. State the quantum numbers I_z , Y and S for the u , d and s quarks and antiquarks. What combination of these leads to the formation of (i) proton and (ii) neutron? (2005)
quarks: $(\frac{1}{2}, \frac{1}{2}, \frac{1}{3})$, $(\frac{1}{2}, \frac{1}{2}, -\frac{1}{3})$, $(\frac{1}{2}, \frac{1}{2}, -\frac{2}{3})$ respectively
proton: uud , neutron: udd
10. What are the basic interactions in nature? Give one example for each. Compare their relative strengths and ranges. (2005)
EM (10^{-2}) / Strong (1) / Grav. (10^{-39}) / Weak (10^{-25})
pairing // strong // decay by weak force // decay by EM, P, n, π^0 // produces
11. Give some characteristic properties of strange particles which distinguish them from non – strange ones. Write the GellMann-Nishijima relation. How is it used for the classification of elementary particles? (2007)
 $Q = I_z + \frac{Y}{2} = I_z + \frac{(B+S)}{2} \Rightarrow S = 2(Q - I_z) - B$
 $\Rightarrow$ strange & non-strange
12. Explain lepton number conservation and why it is necessary to distinguish between different types of neutrinos? (2008)
General of leptons with different mass ($\Sigma L_e = \Sigma L_\mu = \Sigma L_\tau = 0$) $\rightarrow$ to explain why $\pi^- \rightarrow \mu^- + \bar{\nu}_\tau$ don't exist
13. Give the hypercharge and iso spin of the quarks and antiquarks. What are the quark contents of mesons: π^+ , π^0 , π^- , K^+ , K^0 , K^- and η ? (2008)
diff. gen.
quark contents: $u\bar{d}$, $\frac{u\bar{u} - d\bar{d}}{\sqrt{2}}$, $d\bar{u}$, $u\bar{s}$, $\frac{u\bar{u} + d\bar{d} - 2s\bar{s}}{\sqrt{6}}$, $d\bar{s}$, $s\bar{s}$

14. At what energies do we anticipate the unification of strong and electroweak interactions? What will happen to coupling constants of these interactions in this situation? How does unification indicate the decay of proton and existence of lepto quark? (2009)
15. What are the field quanta of weak and electromagnetic interactions? Show that weak interaction is a short range force while electromagnetic interaction is a long range one. (2009)
16. State the quantum number I_z , Y and S for the u d s quarks and anti quarks. Which combination of these leads to the formation of (i) proton and (ii) neutron? (2010)
17. Explain conservation of Baryon Number. Comment on the stability of proton. (2010)
18. Which of the following reactions are permitted or forbidden by various conservation laws?
 (i) $K^- + p \rightarrow \lambda^0 + \pi^0$
 (ii) $K^- + n \rightarrow \Sigma^+ + \pi^0$
 (iii) $\pi^- + p \rightarrow K^- + \Sigma^+$
 (iv) $\pi^+ + n \rightarrow K^- + n^0$
 (v) $\pi^- + p \rightarrow \lambda^0 + K^0$
 (vi) $\pi^- + p \rightarrow K^- + \Sigma^-$ (2012)
19. State the quantum I_z , Y and S for the u d s quarks and antiquarks. Which combination of these leads to the formation of (i) proton and (ii) neutron? (2013)
20. In the following reaction indicate with an explanation, whether they proceed by strong, electromagnetic or weak interaction or they are forbidden:
 (i) $\pi^+ \rightarrow \mu^+ + \nu_\mu$
 (ii) $p \rightarrow n + e^+ + \nu_e$
 (iii) $p + \pi^- \rightarrow K^+ + \Sigma^-$ (2013)
21. (i) What are salient features of nuclear forces?
 (ii) Discuss Yukawa's theory of nuclear forces. (2014)
22. (i) How does liquid drop model explain fission?
 (ii) What are the limitations of shell model? (2014)
23. Describe grand unification theories (GUT). (2015)
24. How many types of neutrinos exist? How do they differ in their masses? (2015)
25. (i) $\Omega^- \rightarrow \Lambda^0 + K^-$
 (ii) $\Lambda^0 \rightarrow p + \pi^-$
 (iii) $K^- \rightarrow \mu^- + \nu_\mu$ (2015)
26. State the quantum numbers I_z , Y and S for u d s quarks and antiquarks. Which combination of these leads to the formation of a (i) proton and (ii) neutron? (2015)

interactions that can occur between the elementary particles.

conservation law, state whether the following reaction are possible or not:

(2016)

Write down the following decays in terms of quarks:

in electroweak unification?

(2016)

from the other.

(2017)

✓ (a) Isotopic spin (strong) & T_3 : strong/EM

(a) Isotopic spin (strong) & T_z : strong/EM

(b) Hyper charge $\text{Strong/EM } (Y = B+S)$

(c) Lepton number (strong, weak, EM)

(d) Charge conjugation (strong/EM)

(2017)

(a) π^+ , (b) K^+ , (c) Δ^{++} , (d) Σ^0 , (e) Ω^-

(2017)

(i) $p + n \rightarrow \Lambda^0 + \Sigma^+$

(ii) $\pi^+ + n \rightarrow \Lambda^0 + K^+$

(iii) $p + n \rightarrow K^+ + \Sigma^+$

(iv) $\pi^0 \rightarrow \gamma + \gamma$

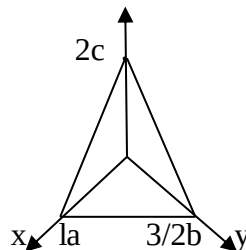
(v) $\bar{n} \rightarrow \bar{p} + e^+ + \nu_e$

(2018)

TUTORIAL SHEET: 8

Crystal Structure & its Determination

1. Obtain reciprocal lattice to normal BCC lattice. (2001)
2. Determine the reciprocal lattice vectors of an fcc lattice. (2005)
3. Show that the reciprocal lattice of a hexagonal lattice is a hexagonal lattice with a rotation of axes
 $\vec{a} = \frac{a}{2}(\sqrt{3}\hat{i} + \hat{j})$, $\vec{b} = \frac{a}{2}(-\sqrt{3}\hat{i} + \hat{j})$, $\vec{c} = c\hat{k}$ (2009)
4. Derive Bragg diffraction condition in vectorial form using incident and diffracted wave vectors and reciprocal lattice vector. $|\vec{G}'| = |\vec{G}' - \vec{K}| = 2\kappa \sin\theta \Rightarrow 2d \sin\theta = n\lambda$ (2010)
5. An electron beam of 4 Kev is diffracted through a Bragg angle of 16° for the first maxima. If the energy is increased to 16 keV, find the corresponding Bragg angle for diffraction. (2010)

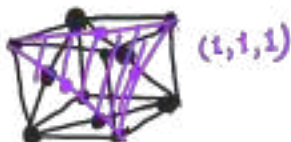


A crystal plane is shown in the above figure. Find its Miller indices and interplanar spacing.

$$\left(\frac{1}{1}, \frac{1}{3/2}, \frac{1}{2}\right) \equiv (6, 4, 3)$$

(2013)

7. Explain the working of SEM and TEM highlight the major differences in principles. Draw neat schematic diagrams. (2014)
8. What is the reciprocal lattice and why is it named so? Derive the relationships for the primitive translation vectors of the reciprocal lattice in terms of those of the direct lattice. $\vec{a}_i \cdot \vec{a}_j = 2\pi\delta_{ij}$ (2014)
9. How does the energy gap in superconductors differ from the energy gap in insulator? How does it vary with temperature for superconductors? $\rightarrow$ formed due to Cooper pairs $\rightarrow$ due to immobility of e^- (2014)
10. Obtain Laue's equations for X-ray diffraction by crystals. Show that these are consistent with the Bragg's law. $\vec{a} \cdot \vec{N} = a \cos\alpha = 2a \sin\theta \cos\alpha$ $\vec{N} = (\hat{i}, -\hat{j})$ $\Delta = \vec{J} \cdot \vec{N}$ (2016)
11. The primitive translation vectors of a two-dimensional lattice are $\vec{a} = 2\hat{i} + \hat{j}$, $\vec{b} = 2\hat{j}$. Determine the primitive translation vectors of its reciprocal lattice. (2016)
12. Deduce the Miller indices of the close-packed planes of atoms in the f.c.c. lattice. (2019)



$$2d \sin\theta = n\lambda$$

$$\cos^2\alpha + \cos^2\beta + \cos^2\gamma = 1$$

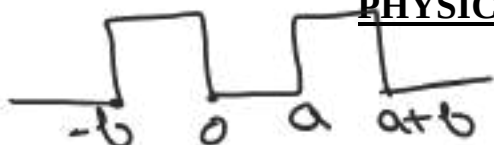
direction cosines of $\vec{N}$ is proportional to $\left\{\frac{a}{h}, \frac{b}{k}, \frac{c}{l}\right\}$

$$\vec{N} \text{ is } \perp \text{ to plane } (h, k, l) \Rightarrow d = \frac{a}{h} \cos\alpha = \frac{b}{k} \cos\beta = \frac{c}{l} \cos\gamma$$

PHYSICS PAPER – II : SOLID STATE PHYSICS

TUTORIAL SHEET: 9

Band theory of solids



1. Show that the condition of periodicity of lattice leads to the Bloch's theorem. What is a Bloch function? (2004)

Handwritten notes for Q1:
 $\psi(x+a) = \psi(x) e^{ik a}$ (under periodic potential)
 $\psi(x) = U(x) e^{ik x}$ (Bloch function)
 $U(x+a) = U(x)$

2. By drawing E vs. $\vec{k}$ curve, distinguish between metal, insulator and semiconductor. (2004)

3. (i) Discuss briefly the concept of effective mass in semiconductors and explain the significance of negative effective mass. (2005)
 (ii) In isolated atoms, the electrons have discrete and definite energies but in solids they have bands of energies. Explain why. (2005)

Handwritten notes for Q3:
 Analogous to concept of $E = \frac{h^2 k^2}{2m^*}$ $\Rightarrow m^* = -\frac{\hbar^2}{\partial^2 E / \partial k^2}$ & $v_g = \frac{dE}{\hbar dk}$
 (Kronig-Penney model) $\hookrightarrow$ constant potential

4. Discuss the motion of an electron in one-dimensional periodic potential and show that it leads to formation of bands of allowed and forbidden states in the electron energy spectrum. How are the Insulators, semiconductors and conductors discriminated on the basis of band structure? (2005)

5. Explain the reduced zone scheme of plotting the energy bands in solids. On the basis of E vs k curve distinguish between conductors, semiconductors and insulators. (2010)

6. What is nearly free electron approximation? On the basis of this approximation explain the formation of energy bands in solids. (2010)

Handwritten notes for Q6:
 $P = \frac{m v_{gab}}{\hbar^2} = \frac{P_{gab}}{\hbar^2}$
 $-1 \leq \frac{P_{gab}}{\alpha a} + \cos \alpha a \leq 1$ $\left(\alpha = \sqrt{\frac{2mE}{\hbar^2}} \right)$

7. Starting with the expression for the density of states for electrons in a band, show that the Fermi energy of an intrinsic semiconductor is at the middle of the band gap. Use these results to estimate the electron density at 300K (Assuming $E_g = 1$ eV and the rest masses of electron and hole as m_e and m_h). (2013)

8. Describe the motion of an electron in one dimensional periodic potential and show that it leads to formation of bands of allowed and forbidden states in the electron energy spectrum. How are the conductors, semiconductors and insulators discriminated on the basis of band structure? (2015)

9. What is the difference between direct and indirect band gap semiconductors? Which one is suitable for use in solar cells? (2016)

10. The Energy (E) and wave vector (k) for a conduction band electron in a semiconductor are related as $E = \alpha \frac{\hbar^2 k^2}{m_0}$ where α is a constant and m_0 is the free electron mass. Calculate the effective mass of the electron. (2017)

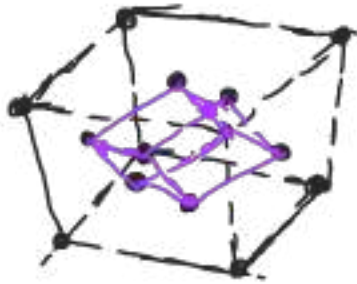
Handwritten note for Q10:
 $m^* = \frac{\hbar^2}{\partial^2 E / \partial k^2}$

11. Consider a face-centred cubic lattice of side a. Deduce –

- (i) The primitive translation vectors; $\rightarrow$

- (ii) The volume of the primitive cell;
- (iii) The reciprocal primitive translation vectors;
- (iv) The volume of the reciprocal lattice

(2019)



→ Rhombic dodecahedron

$$\vec{a} = \frac{a}{2}(\hat{j} + \hat{k}) \quad \text{[FCC]}$$

$$\vec{b} = \frac{a}{2}(\hat{k} + \hat{i})$$

$$\vec{c} = \frac{a}{2}(\hat{i} + \hat{j})$$

TUTORIAL SHEET: 10 Thermal and Magnetic Properties of Solids

- ✓ 1. Distinguish between Einstein and Debye models of specific heats of solids. Obtain an expression for the specific heat of a solid on the basis of Debye model. Discuss the results for low and high temperature ranges. (2010)
2. Distinguish between classical and quantum theory of paramagnetism. Explain the paramagnetism of free electrons on the basis of F.D. distribution. (2010)
- ✓ 3. Find an expression for lattice specific heat of solids, and its low and high temperature limits. What is Debye temperature? (2010)
- ✓ 4. Find an expression for lattice specific heat of solids and its low and high temperature limits. What is Debye's temperature? (2011)
- ✓ 5. The velocity of sound in f.c.c. gold and f.c.c. copper is 2100m/s and 3800 m/s respectively. If the Debye temperature of copper is 348 K, then determine the Debye temperature of gold. Take the densities of gold and copper as $1.93 \times 10^4 \text{ kg/m}^3$ and $0.89 \times 10^4 \text{ kg/m}^3$ respectively. (2014)
- ✓ 6. Find an expression for lattice specific heat of a solid, and its low and high temperature limits. What is Debye temperature? (2015)
- ✓ 7. Write down the salient features of the Einstein's theory of lattice heat capacity. Further write down the expression for specific heat in Einstein's theory and explain its high and low temperature limits. (2016)
- ✓ 8. In a semiconductor, the effective masses of an electron and hole are $0.07 m_0$ and $0.4 m_0$ respectively, where m_0 is the free electron mass. Assuming that the average relaxation time for the hole is half of that for the electrons, calculate the mobility of the holes when the mobility of the electrons is $0.8 \text{ m}^2 \text{ volt}^{-1} \text{ s}^{-1}$. $\sigma = \frac{ne^2\tau}{m^*} = ne\mu \Rightarrow \mu = \frac{e\tau}{m^*}$ (2017)
- ✓ 9. Derive an expression for lattice specific heat in Debye model. Find its low temperature limit (Debye T^3 law). (2018)

⑤ $\left[\frac{3}{D} = \frac{9N}{4\pi V} \left(\frac{2}{v_1^3} + \frac{1}{v_2^3} \right) \right]^{-1}$; for fcc $\Rightarrow v_1 = v_2$

$\left\{ \begin{array}{l} M_{\text{gold}} = 197 \text{ gram} \text{ \& } M_{\text{copper}} = 63.55 \text{ gram} \\ \rho_{\text{gold}} = \frac{M_{\text{gold}}}{V_1} \\ \rho_{\text{copper}} = \frac{M_{\text{copper}}}{V_2} \end{array} \right. \quad \left(\frac{v_1}{v_2} \right)^3 = \left[\left(\frac{\rho_1}{\rho_2} \right)^3 = \frac{V_2 v_2^3}{V_1 v_1^3} \right]$

$\Rightarrow \frac{V_2}{V_1} = \frac{\rho_1}{\rho_2} \times \frac{M_2}{M_1}$

SOLID STATE PHYSICS TUTORIAL SHEET: 11 SUPERCONDUCTIVITY

1. Describe Josephson junction and the effect. How is it exploited in the formation of a SQUID? Write down the applications of a SQUID?
e⁻ - phonon - e⁻ interaction
magnetic levitation, Hyperloop, better superconductor. (2001)
2. Give a qualitative account of BCS theory of superconductivity.
T_c is quite high
H₂S (2015) ⇒ T = 203 K (under high P)
YBCO, TBCO (T_c = 125 K) (2002)
3. Explain Meissner effect and Josephson effect. Describe their applications in the field of superconductivity.
T_c is quite high
High T ⇒ high studies
Low T ⇒ Cooper pairs
High T ⇒ commercially useful
can't be explained by BCS theory (2003)
4. What are high temperature superconductors? Give two examples indicating their transition temperature. Differentiate between the conventional and high temperature superconductivity.
T_c is quite high
High T ⇒ high studies
Low T ⇒ Cooper pairs
High T ⇒ commercially useful
can't be explained by BCS theory (2003)
5. What is a Josephson junction? Discuss DC and AC Josephson effects and obtain the relation $v = 2eV/h$, where symbols have their usual meanings. (2003)
6. What is Josephson Effect? Discuss briefly DC and AC effects. Give some practical applications of Josephson junctions.
measurement of v/e
ω = 2eV/h ⇒ high ω for V₀ = 1 μV
SQUID
Sensitive magnetometer (2007)
7. Distinguish between 'soft' and 'hard' superconductors. Explain how penetration depth varies with magnetic field strength and temperature.
[1/λ₀² = 1/λ₀² (1 - (T/T_c)⁴)]
[H_c(T) = H_c(0) (1 - (T/T_c)²)] (2008)
8. A type-II superconducting material with superconducting transition temperature T_c is placed in a magnetic field. What is the material state (normal or superconducting or mixed) for
(i) $T < T_c; H > H_{c_2}$ (Normal) (ii) $T > T_c; H < H_{c_1}$ (Normal)
(iii) $T < T_c; H_{c_1} < H < H_{c_2}$ (Mixed) (iv) $T > T_c; H > H_{c_2}$ (Normal) (2009)
9. Describe the characteristic properties of a superconductor. Derive London equation for a superconductor and hence explain Meissner effect. (2010)
10. What is Josephson tunneling? Distinguish between a.c. and d.c. Josephson effects.
constant V₀
Quantum tunneling (2010)
11. Describe the Characteristics properties of a superconductor. Derive London equation for a superconductor and hence explain Meissner effect.
diamagnetic, repels B, meissner effect, Cooper pairs, SQUID (quantum tunneling of flux) (2011)
12. Explain the variation of Magnetisation of Type I and Type II superconductors as a function of applied magnetic field
(sudden loss)
(gradual loss of P) (2012)
13. Cooled to 4 K and then magnetic field is applied. With schematic diagrams, explain path of magnetic field lines in all these situations:
(from T < T_c) (2013)
14. Lead in the superconducting state has critical temperature of 6.2 K at zero magnetic field and a critical field of 0.064 T at 0 K. Determine the critical field at 4 K.
T_c = 6.2 K
H_c = H_c(0) (1 - (T/T_c)²)
= 0.064 T (1 - (4/6.2)²) = 0.037 T (2014)

- ✓ 15. Distinguish between a superconductor and perfect conductor. Explain what is a Cooper pair. (2015)
Cooper pair mesomer effect
- ✓ 16. With the help of a schematic diagram, show how entropy and specific heat vary with temperature for a superconductor. (2016)
Le⁻ - phonon - e⁻ pair
 $\left\{ \vec{\nabla} \times \vec{J} = -\frac{ne^2}{m} \vec{B} \right\}$ $\left[\frac{\partial \vec{J}}{\partial t} = \frac{\vec{E}}{m_0 \gamma} \right]$ & $\vec{\nabla} \times \vec{B} = \mu_0 \vec{J}$ & $\vec{\nabla} \times (\mu_0 \vec{J}) = -\nabla^2 \vec{B}$
- ✓ 17. Obtain the expression for penetration depth using London's equation of superconductivity and explain its significance. (2017)
very small for soft superconductor
characteristic λ_L is distance upto which $\vec{B}$ can penetrate $\rightarrow$ mesomer effect
- ✓ 18. What are type I and type II superconductors? Give examples. Discuss and compare Meissner effect and perfect diamagnetic behaviors for type I and type II superconductors. (2018)

For lattice constant a ,

$$a^3 = \frac{n_{\text{eff}} \cdot M_{\text{molar}}}{N_A \rho}$$

$$n_{\text{eff}} \rightarrow a^3$$

$$\frac{n_{\text{eff}} M_{\text{molar}}}{N_A} \rightarrow a^3$$

$$\rho = \frac{\text{mass}}{\text{volume}}$$

&

$$\text{number density (n)} \times \text{mass} = \text{density (}\rho\text{)}$$

$$\{eD = \mu k_B T\}$$

$$V_{bi} = \frac{k_B T}{e} \ln\left(\frac{n_0}{n_i}\right)$$

$$\Delta p = \Delta p_0 \exp\left(-\frac{L}{\sqrt{D\tau}}\right)$$

$$\left[\text{Conduction} = \frac{e}{4\pi n_i e (\mu_n + \mu_p)} \right]$$

Electronics/ TUTORIAL SHEET: 12 Semiconductors/Solar cell



1. Draw the device structure of a p-n junction solar cell and explain how light energy is converted into electrical energy. Draw and explain its I-V characteristics. (2015)
2. A silicon semiconductor sample at $T = 300 \text{ K}$ having cross-sectional area of $0.5 \mu\text{m}^2$ has a pentavalent donor doping profile given by $(x) = 5 \times 10^{16} e^{(-x/L_n)} \text{ cm}^{-3}$. Given, the mobility of the electrons in the sample is $1250 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ and the diffusion length of the electrons, L_n , is $4 \mu\text{m}$. Calculate the diffusion current in the sample at distance $x = 2 \mu\text{m}$. (2019)
 $V_T = 0.026 \text{ Volt}$
 $D_p = \mu_p V_T$
 $J_p = e D_p n_0 \frac{dn}{dx}$
 $n = \frac{n_i^2}{p}$
 $E_g = E_g - k_B T \ln\left(\frac{p}{n_i}\right)$
3. A silicon semiconductor sample is doped with $6 \times 10^{16} \text{ cm}^{-3}$ of aluminium and $7 \times 10^{15} \text{ cm}^{-3}$ of phosphorus atoms. Given at $T = 300 \text{ K}$, the intrinsic carrier concentration, $n_i = 1.5 \times 10^{10} \text{ cm}^{-3}$; the band gap, $E_g = 1.1 \text{ eV}$; the electron mobility, $\mu_n = 1250 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ and the hole mobility, $\mu_p = 480 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$. Determine in the sample of the following:
 (i) The type of the semiconductor, n or p
 (ii) The hole carrier concentration
 (iii) The electron carrier concentration
 (iv) The position of the Fermi level in the sample with respect to the bottom of the conduction band
 (v) The conductivity of the sample
 $\{N_A > N_D\} \Rightarrow \text{p-type}$
 $p \sim (N_A - N_D)$
 $n = \frac{n_i^2}{p}$
 $E_F = E_g - k_B T \ln\left(\frac{p}{n_i}\right)$
 E_F is at mid of band gap
 $\sigma \sim p N_A - \mu_p$
4. A 5 cm^2 Ge solar cell with a dark reverse saturation current of 2 nA has solar radiation incident upon it, producing 4×10^{17} electron-hole pairs per second. The electron and hole diffusion lengths are given to be $5 \mu\text{m}$ and $2 \mu\text{m}$, respectively. Calculate for the cell of the following:
 (i) The short-circuit current
 (ii) The open-circuit voltage
 $p = n_i \exp\left(-\frac{E_F - E_i}{k_B T}\right)$

$$\left[\text{Diffusion length, } l_p = \sqrt{D_p \tau_p} \right]$$

Hall effect is appearance of $\vec{E}$ in conductor due to current & $\vec{B}$ in 1^{st} direction

$$eE_H = e v B$$

$$\& J = n_0 e v$$

$$R_H = \frac{E_H}{J B} = \frac{1}{n_0 e}$$

for BCC, 2 atoms $\rightarrow a^3$
 $n = \frac{2}{a^3}$

Solar cells are photovoltaic cells converting solar radiation incident into electrical energy.

For a solar cell, I-V characteristic equation

$$I = -I_{ph} + I_0 \left(\exp\left(\frac{qV}{k_B T}\right) - 1 \right)$$

From open circuit, $I = 0$ & assuming $V_{oc} \gg \frac{k_B T}{q}$

$$V_{oc} = \frac{k_B T}{q} \ln\left(\frac{I_{ph}}{I_0}\right) \quad (1)$$

Since $I_{ph} = I_{max}$ (at $I = I_{ph}/V_{oc}$)

Since I_{ph} is light intensity, thus,

$$I_{ph2} = I_{ph1} \times 2 \quad (\text{as light intensity is doubled})$$

$$\Rightarrow I_{ph2} = 300 \text{ mA} \quad (\text{as } I_{ph1} = I_{sc} = 150 \text{ mA})$$

Similarly from equation (1),

$$V_{oc2} - V_{oc1} = \frac{k_B T}{q} \ln\left(\frac{I_{ph2}}{I_{ph1}}\right)$$

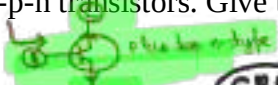
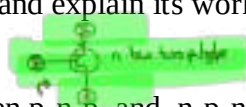
$$p = n_i \exp\left(-\frac{E_F - E_i}{k_B T}\right) \quad (2019)$$

TUTORIAL SHEET: 13

TRANSISTORS (Amplifiers/Oscillators)

1. Circuit as shown in the figure. It is designed to establish the quiescent potential at $V_{CE} = 12V$, $I_C = 1.5 \text{ mA}$ and stability factor $S \leq 3$. Find the values of resistors R_E , R_1 and R_2 where symbols have their usual meaning. (2004)
2. Explain the working of a phase-shift oscillator. Draw the circuit diagram of a phase-shift oscillator using either a transistor or an op-amp. What are the conditions which must be satisfied to achieve stable oscillations? Calculate the frequency of oscillations for $R = 1k\Omega$ and $C = 01 \mu F$. (2007)
3. Explain the difference between n-p-n and p-n-p transistors. Give their device structure and biasing circuits. Draw a circuit for a single-stage common-emitter amplifier. (2007)
4. Draw the circuit diagram of a two-stage RC-coupled common emitter transistor amplifier. Show how the magnitude and phase of voltage gain vary with frequency. Define bandwidth of this amplifier. An amplifier with open loop voltage gain, $A_v = 1000 \pm 100$ is available. It is necessary to have an amplifier whose voltage gain varies by no more than $\pm 0.1\%$. Find the reverse transmission factor β of the feedback network used and the gain with feedback. (2008)
5. A certain colpits oscillator uses a tank circuit with $L = 20\text{mH}$; $C_1 = 200 \text{ pf}$ and $C_2 = 200 \text{ pf}$. What is the frequency of oscillation? (2008)
6. (i) "Transistors are current operated devices, while vacuum tubes are voltage operated" – Explain.
(ii) Why are junction transistors called bipolar devices? (2008)
7. Draw the common-base amplifier circuit, using an n-p-n transistor and briefly discuss its working. (2010)
8. Draw and explain the collector characteristics of a bipolar junction transistor in common emitter configuration. Using the plot, explain how the transistor can be used as an ON-OFF switch. (2013)
9. Explain how the circuit shown above can be a source of oscillations. Use this circuit to construct a transistor oscillator and explain its working. What is the frequency of oscillations of this circuit? (2013)
10. Differentiate between p-n-p and n-p-n transistors. Give their device structure and biasing circuits when used as an amplifier. (2015)
11. Design a transistor based Colpitts oscillator which can oscillate at 9 MHz. Explain how the oscillations are created and sustained. (2015)

(EBC)



(CBE)

Feedback for active device is taken from voltage divider with Tank circuit.

$$L_{eq} = L_1 + L_2 + 2M$$

40

$$f_{Hartley} = \frac{1}{2\pi\sqrt{L_{eq}C}}$$

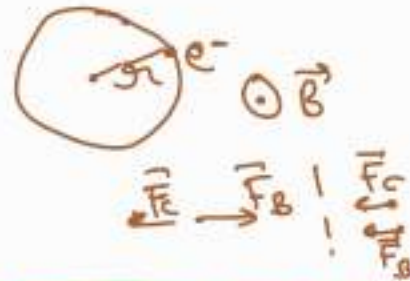
TUTORIAL SHEET: 14 Digital Electronics/Microprocessors

- ✓ 1. Simplify the equation $Y = [A\bar{B}(C + BD) + \bar{A}BC]$ Using Boolean algebra. Give the logic circuit for the equation before and after the simplification. (2007)
- ✓ 2. Give the Boolean expression and the truth table for an XOR gate. Why are NAND gates called universal gates? Realise OR and XOR gates employing only NAND gates. (2007)
- ✓ 3. How can the NAND-gates be combined to perform the OR – operation? (2008)
 $Y = A + B = \overline{\overline{A} \cdot \overline{B}}$ 
- ✓ 4. Simplify the logical expression $AB + AB + ABC$ using a karnaugh map. (2010)
 $\hookrightarrow \begin{array}{cc|cc|cc|cc} & 0 & 1 & 0 & 1 & 0 & 1 & 0 & 1 \\ \hline 0 & 1 & 1 & 1 & 1 & 1 & 1 & 1 & 1 \\ 1 & 1 & 1 & 1 & 1 & 1 & 1 & 1 & 1 \end{array} \quad Y = ABZ + ABC = AB$
- ✓ 5. Simplify the logical expression $(A + B)(B + C)(A + C)$ and draw the logical circuit to implement it. (2010)
 $Y = (B + AC)(A + C) = AB + BC + AC$
- ✓ 6. Construct a digital circuit to add three bits A, B and C and provide their sum and carry as-outputs. Show appropriate Boolean expressions and truth table to justify the outputs. (2013)
 $\text{Carry} = (A \oplus B) \cdot C_{in} + AB$ $\rightarrow \overline{AB} = \overline{A} + \overline{B}$ & $\overline{(A+B)} = \overline{A} \cdot \overline{B}$
 $\hookrightarrow S = A \oplus B \text{ \& } C = AB$ → Full Adder
- ✓ 7. (a) Give the two forms of the De Morgan's theorem. $\rightarrow \overline{AB} = \overline{A} + \overline{B}$ & $\overline{(A+B)} = \overline{A} \cdot \overline{B}$
(b) What is the principle used in 'half-adder'? Give the truth table for it. (2014)
- ✓ 8. Simplify the logical expression $[AB(C + BD) + A B] C$. (2015)
- ✓ 9. Why are NAND and NOR gates called universal gates? Give the logic diagram, Boolean equation and the truth table of NAND gate. (2016)

$$\chi_{\text{diamagnetic}} = -\frac{N\mu_0 Z e^2}{6m} \langle r^2 \rangle$$

Don't forget to see examples of Puri & Babbar

Classical Zeeman effect (Mains 2020)



$$m\omega^2 \pm eVB = m\omega(\omega + d\omega)^2$$

$$m\omega^2 \pm eVB = 2m\omega d\omega + m\omega^2 d\omega^2$$

(neglecting $(d\omega)^2$)

$$\Rightarrow 2m\omega d\omega = \pm eVB$$

$$\{v = \omega\lambda\}$$

$$\Rightarrow d\omega = \pm \frac{eB}{2m}$$

$$\left\{ \omega = \frac{2\pi c}{\lambda} \Rightarrow d\omega = -\frac{2\pi c}{\lambda^2} d\lambda \right\}$$

$$\Rightarrow -\frac{2\pi c}{\lambda^2} d\lambda = \pm \frac{eB}{2m}$$

$$\Rightarrow \frac{e}{m} = \pm \frac{4\pi c}{B\lambda^2} d\lambda$$

$$\left\{ d\lambda = \pm \frac{Be\lambda^2}{4\pi mc} \right\}$$

$$\left\{ \nu = \frac{c}{\lambda} \Rightarrow d\nu = -\frac{c}{\lambda^2} d\lambda \right\}$$

or,

$$d\nu = \pm \frac{eB}{4\pi m}$$

→ required expression



→ If we observe spectra parallel to applied $\vec{B}$ → Longitudinal → only 2 lines

Transverse → 3rd line